

NPAC

Noyaux
Particules
Astroparticules
Cosmologie

Master 2 Recherche

Bruno Mazoyer - LAT Orsay

Detector physics – NPAC 2022-2023

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Plan

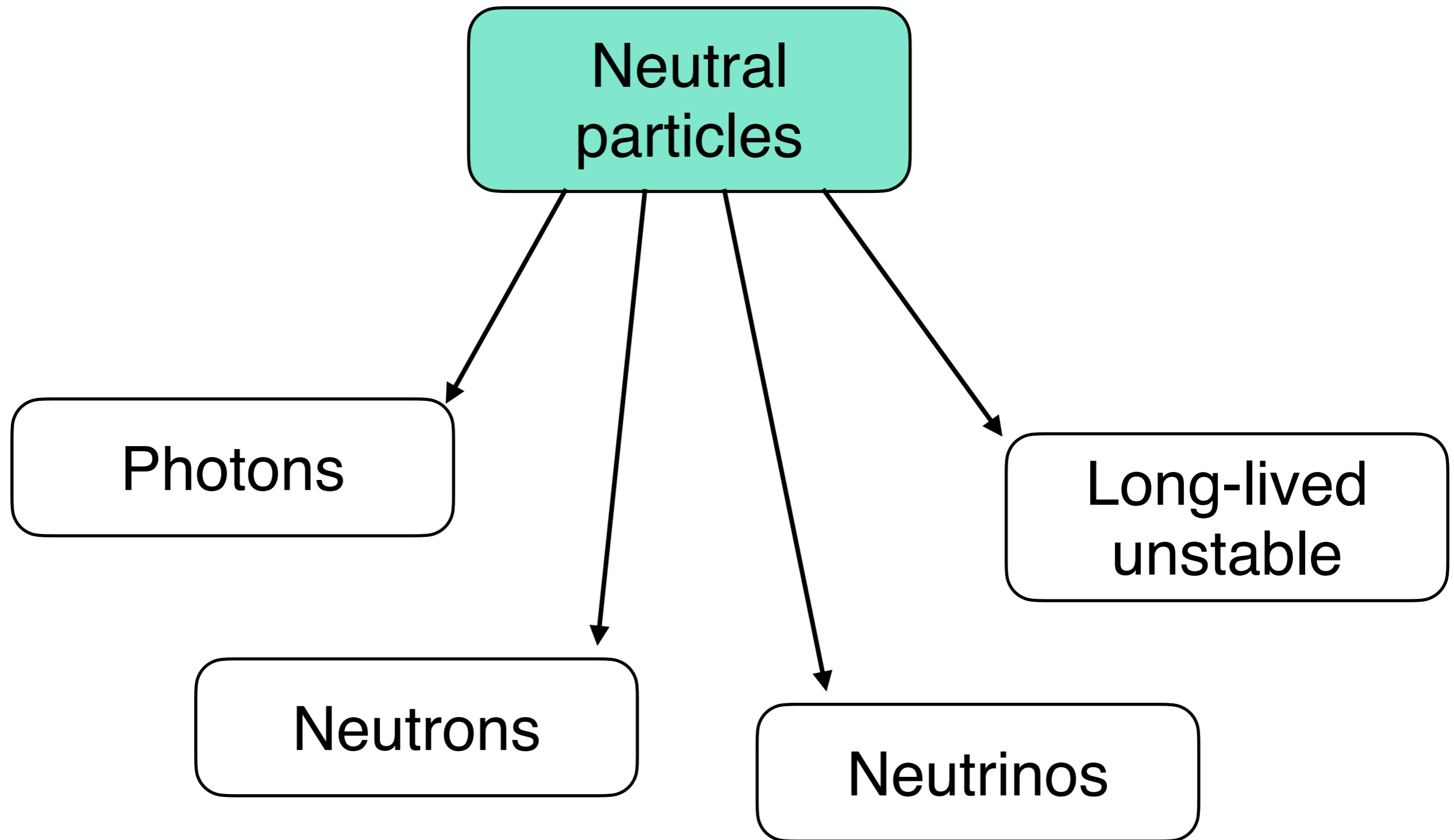
- Second lecture: intro, interaction of charged particles in matter
- Third lecture (this one):
 - Interactions of photons in matter
 - Start of scintillators & photodetectors
- Fourth lecture: photodetectors, interactions of other neutral particles in matter

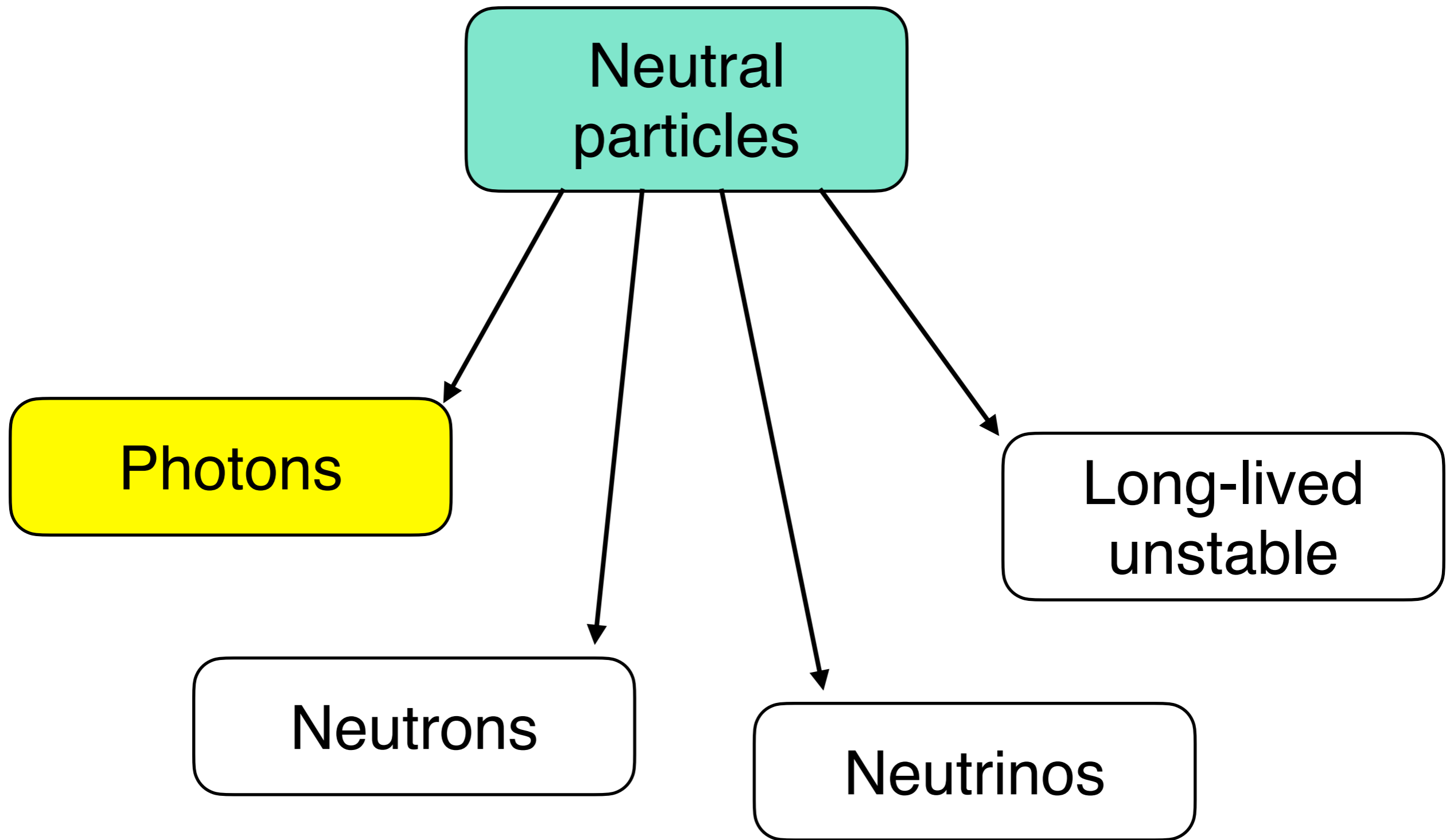
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Interactions of neutral particles in matter





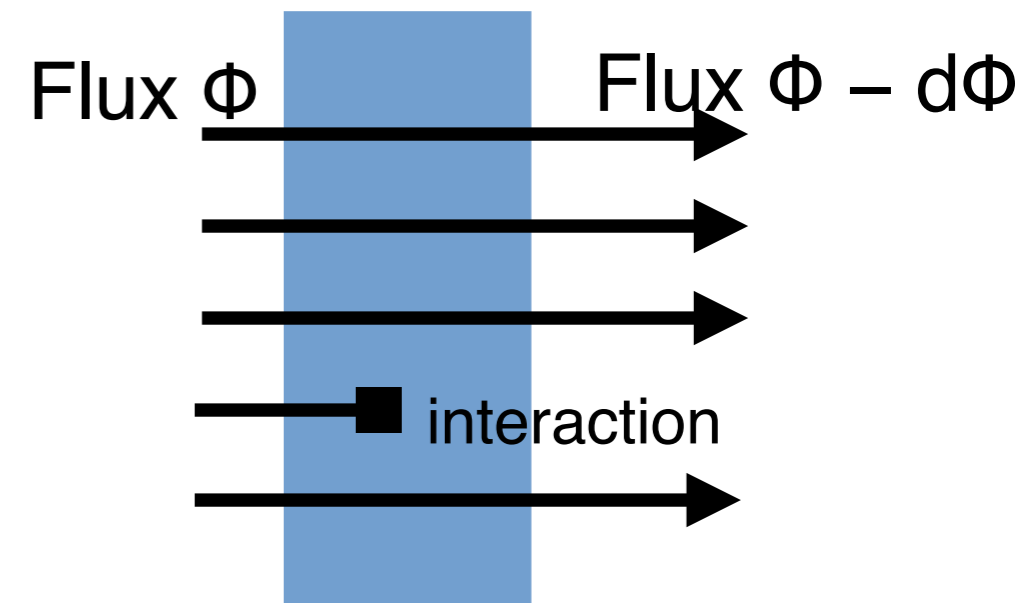
Interaction of photons with matter

- Principal characteristic:
A single interaction removes photon from beam (not the case for heavy charged particles)
- Possible interactions:
 - Photoelectric effect ($\gamma A \rightarrow A^+ e^-$; $A = \text{atom}$)
 - Compton scattering ($\gamma e \rightarrow \gamma e$; inelastic)
 - Pair production ($\gamma N \rightarrow e^+ e^- N$; $N = \text{nucleus}$)
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 - ...



with $d\Phi = \mu \Phi dx$,

where μ is the absorption coefficient and depends on E , Z , and target density n [with $n = \rho N_A / A$ for atoms]

You can show $d\Phi = -\Phi n \sigma dx$, and thus the mean free path is

$$\lambda = 1/\mu = 1 / (n\sigma)$$

for total absorption cross-section σ

Interaction of photons with matter

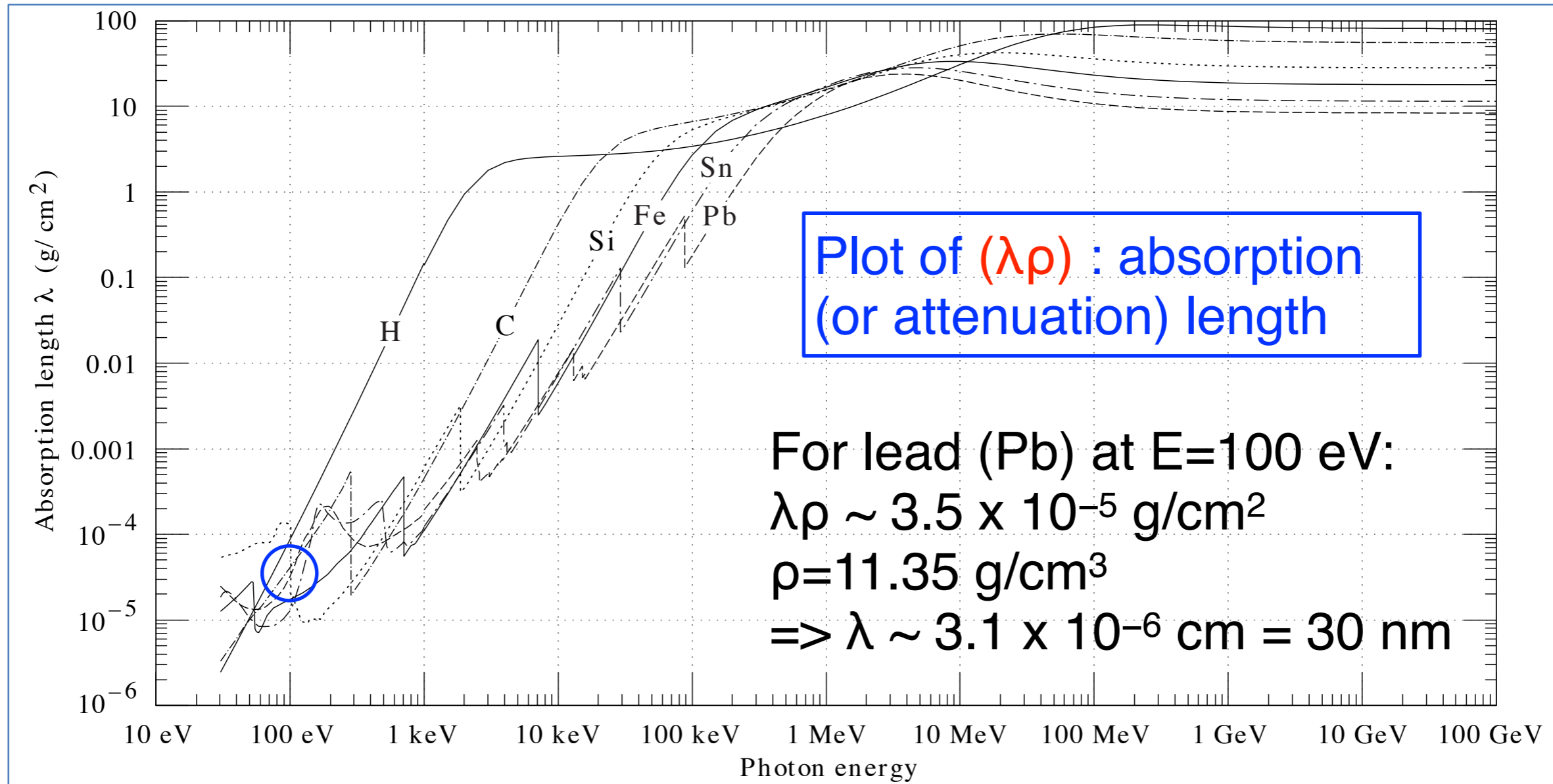


Figure from PDG: K.A. Olive et al. (Particle Data Group), Chin. Phys. C, 38, 090001 (2014) and 2015 update

- Interaction probability of low-energy photons is very high (30nm in lead at 100 eV)
- Peaks are observed
- Saturation at high energy

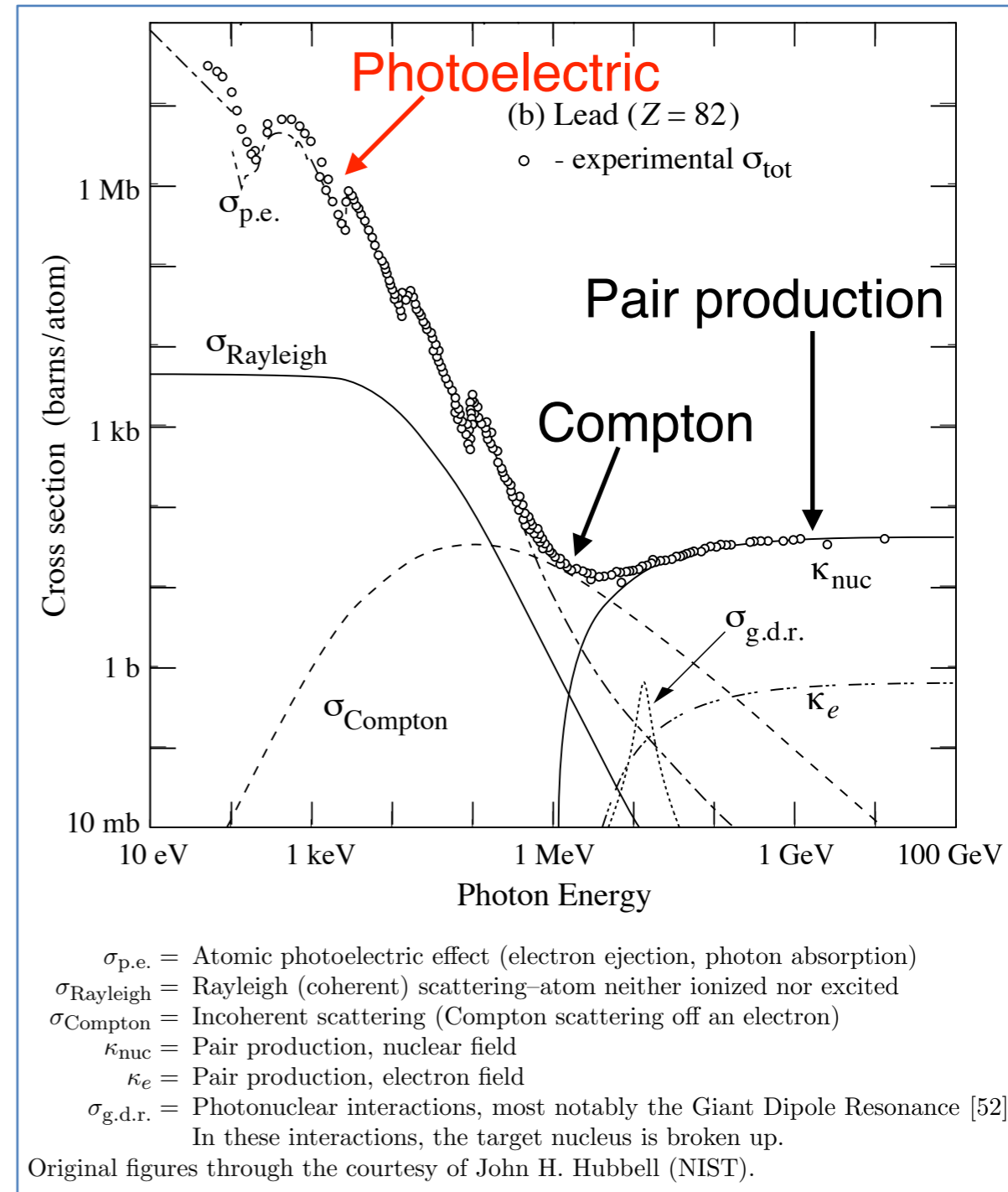
Caution: watch out for notation difference here (our mean free path λ in cm).

Interaction of photons with matter

The dominant interaction is one of these three for photon energies above 10 eV.

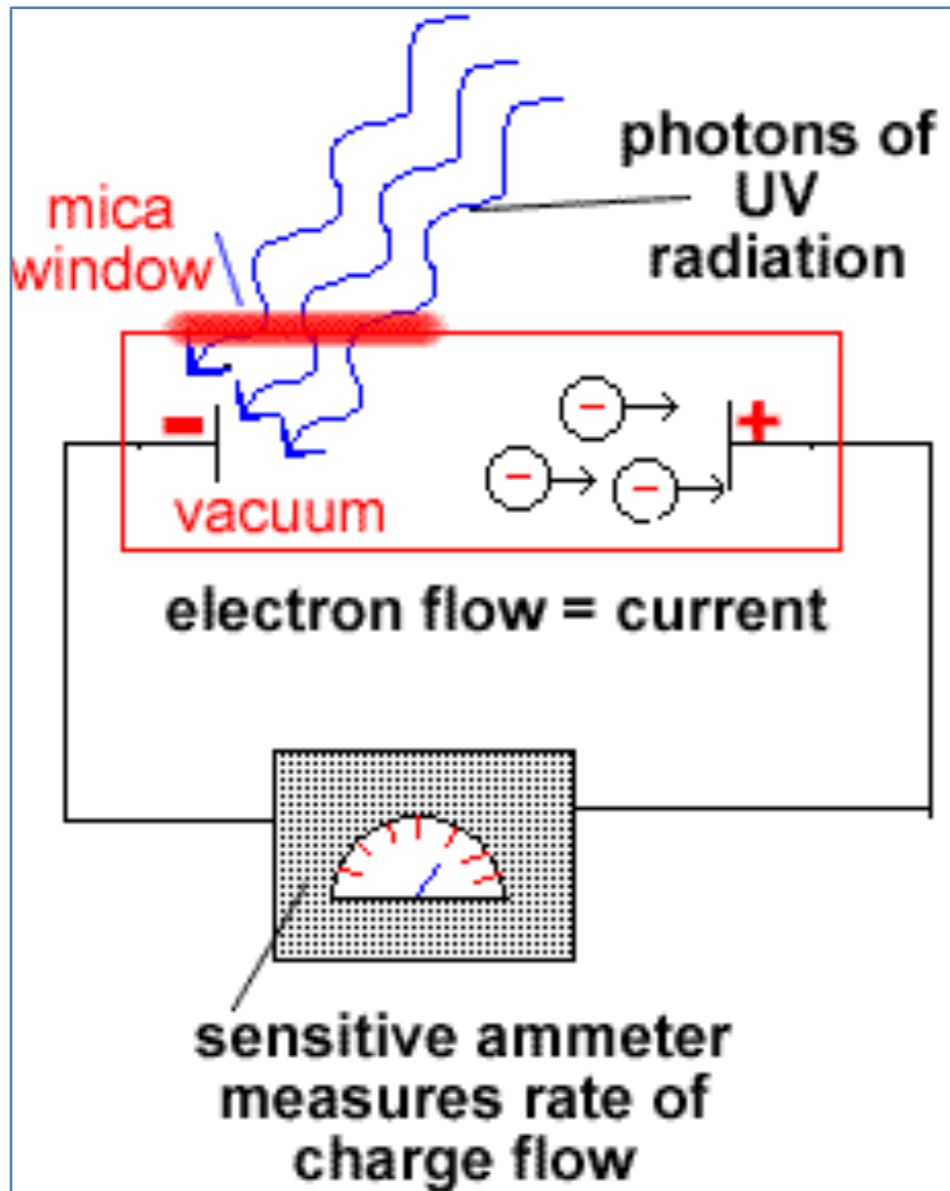
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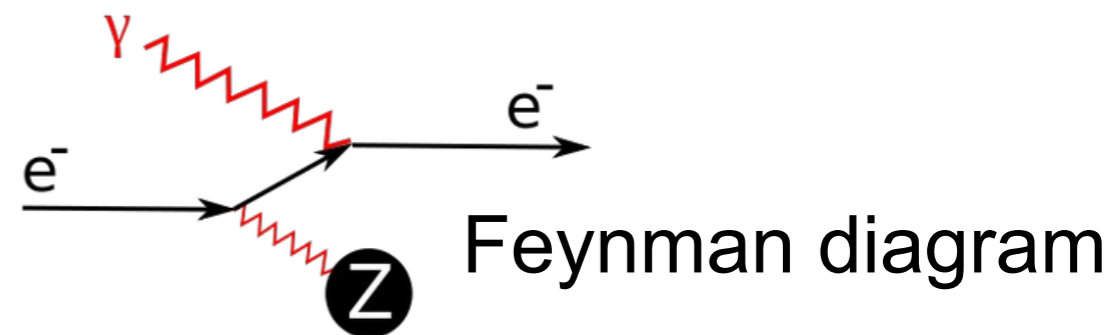
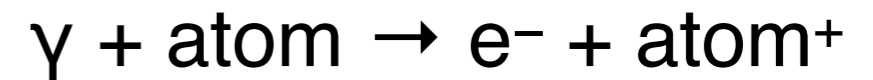
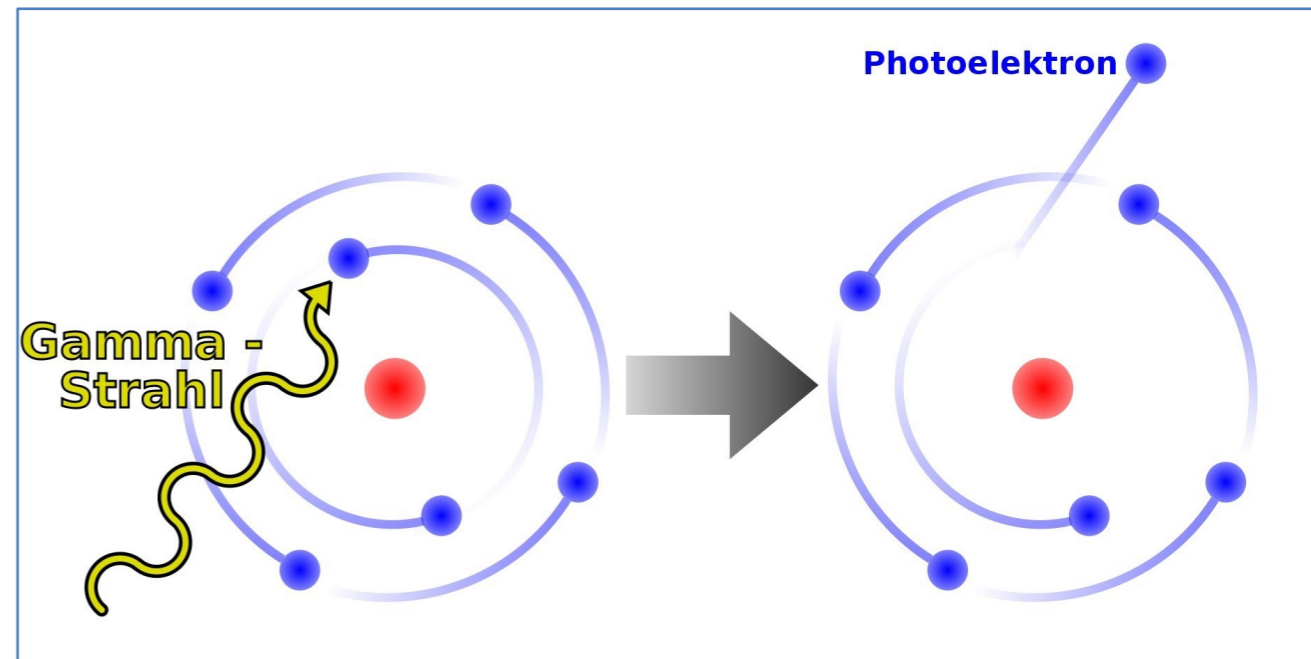


Photons & matter: Photoelectric Effect

Discovered before 1900.



Naïve picture:



Q: Why is the reaction $\gamma e^- \rightarrow e^-$ impossible? What is the role of the nucleus in the photoelectric effect?

Photons & matter: Photoelectric Effect

Kinetic energy of outgoing electron:

$$T_e = h\nu - I_b$$

Photon energy E_γ

Binding energy

Typical energy dependence:

$$\sigma_{\text{ph}} = 2\pi r_e^2 \alpha^4 Z^5 mc^2 / E_\gamma \quad \text{for } E_\gamma \gg mc^2$$

$$\sigma_{\text{ph}} = \pi \alpha r_B Z^5 (I_b / E_\gamma)^{7/2} \quad \text{for } I_b \ll E_\gamma \ll mc^2$$

Example:

$$r_B = 0.53 \times 10^{-10} \text{ m}$$

$$I_b = 13.6 \text{ eV}$$

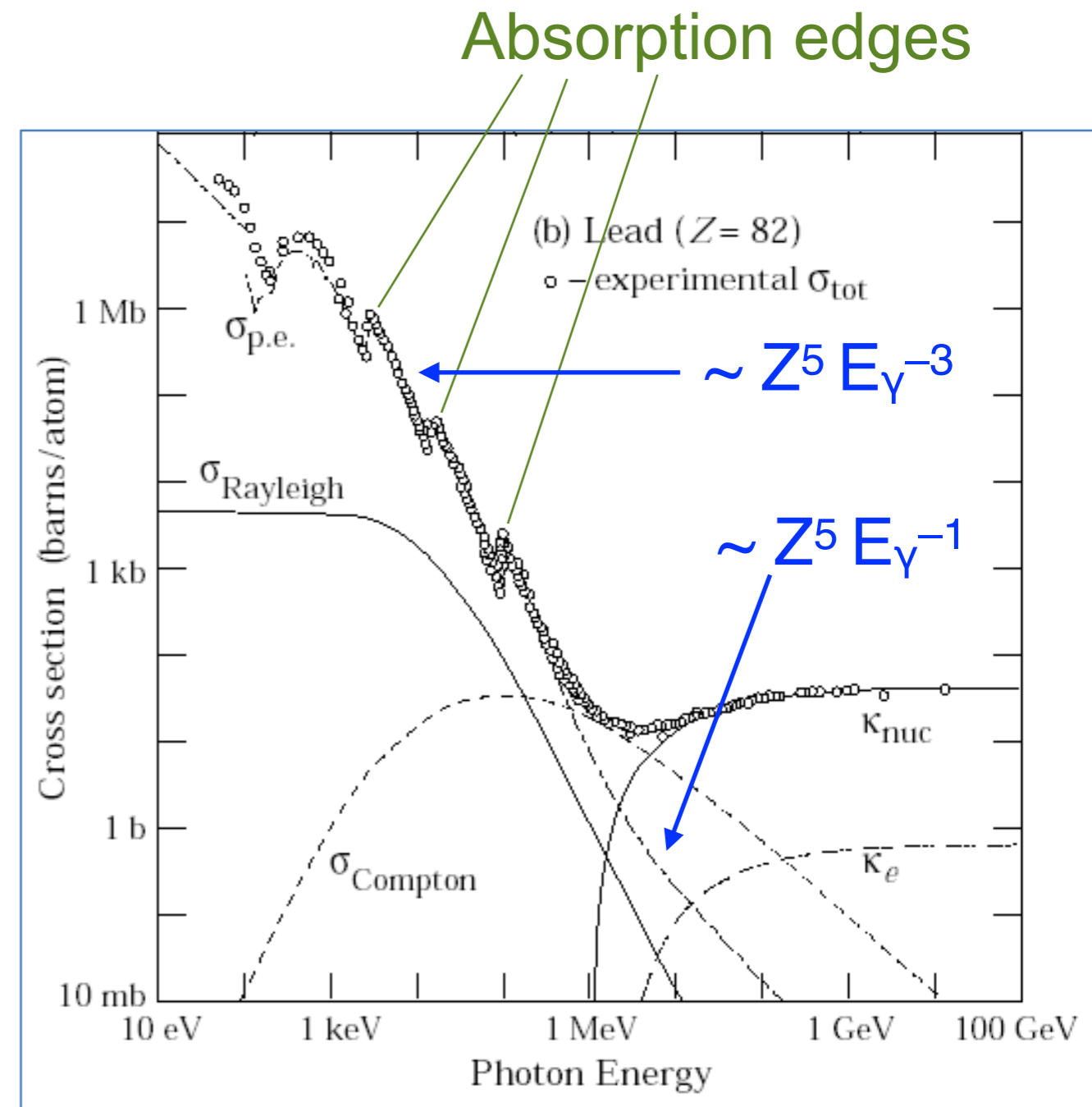
$$E_\gamma = 100 \text{ keV} :$$

$$\sigma_{\text{ph}}(\text{Fe}) = 29 \text{ barn}$$

$$\sigma_{\text{ph}}(\text{Pb}) = 5000 \text{ barn}$$

Strong dependence on Z !

(NB: 1b = 1 barn = $10^{-24} \text{ cm}^2 = 10^{-28} \text{ m}^2$)



Photons & matter: Photoelectric Effect

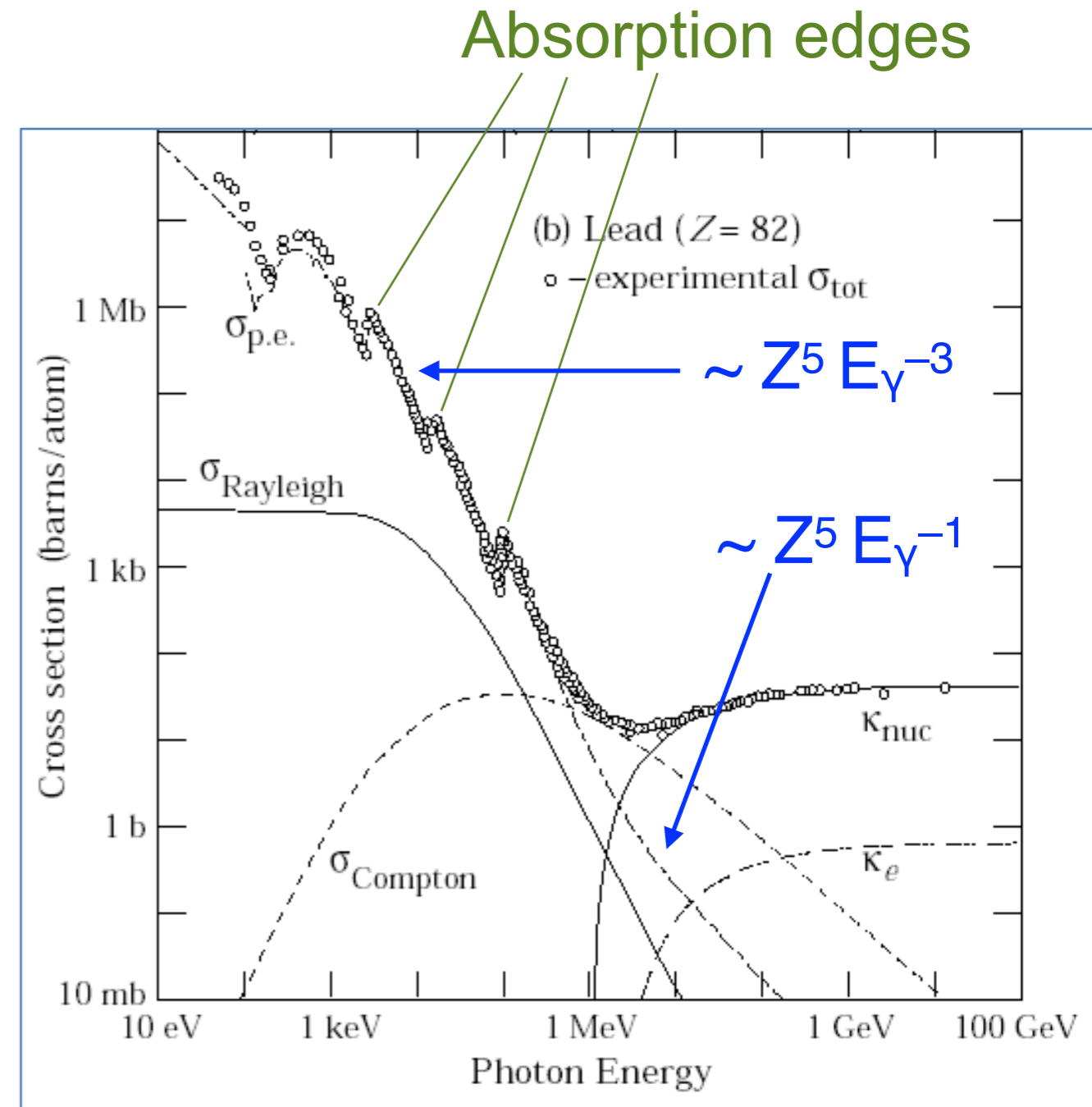
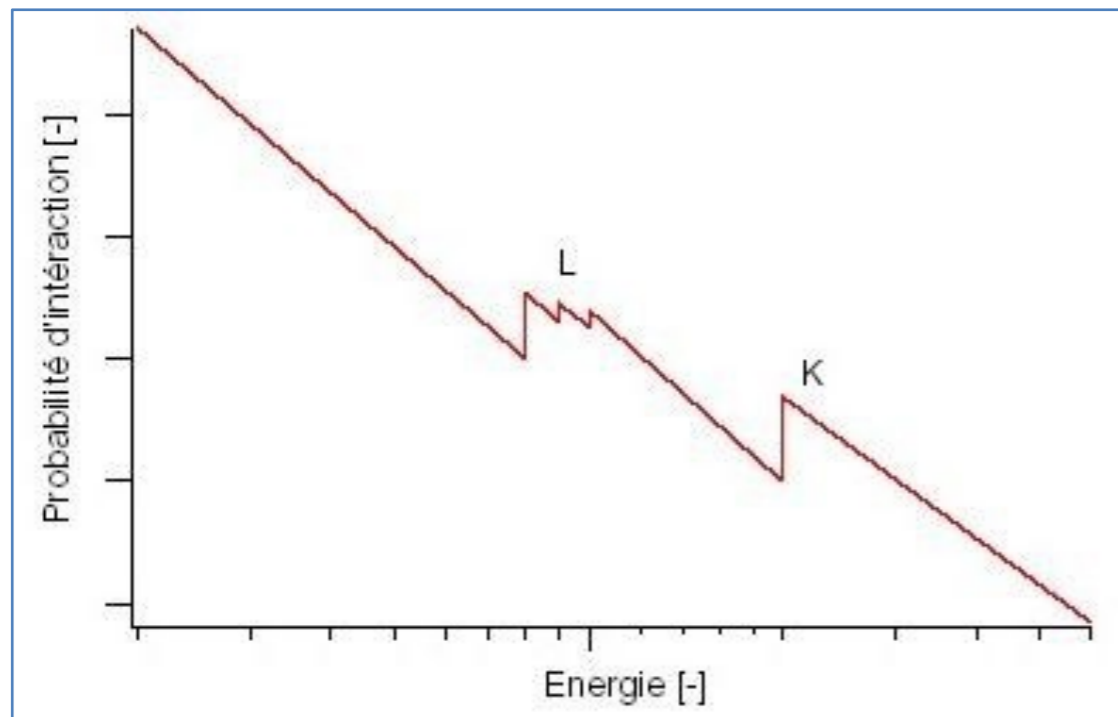
Kinetic energy of outgoing electron:

$$T_e = h\nu - I_b$$

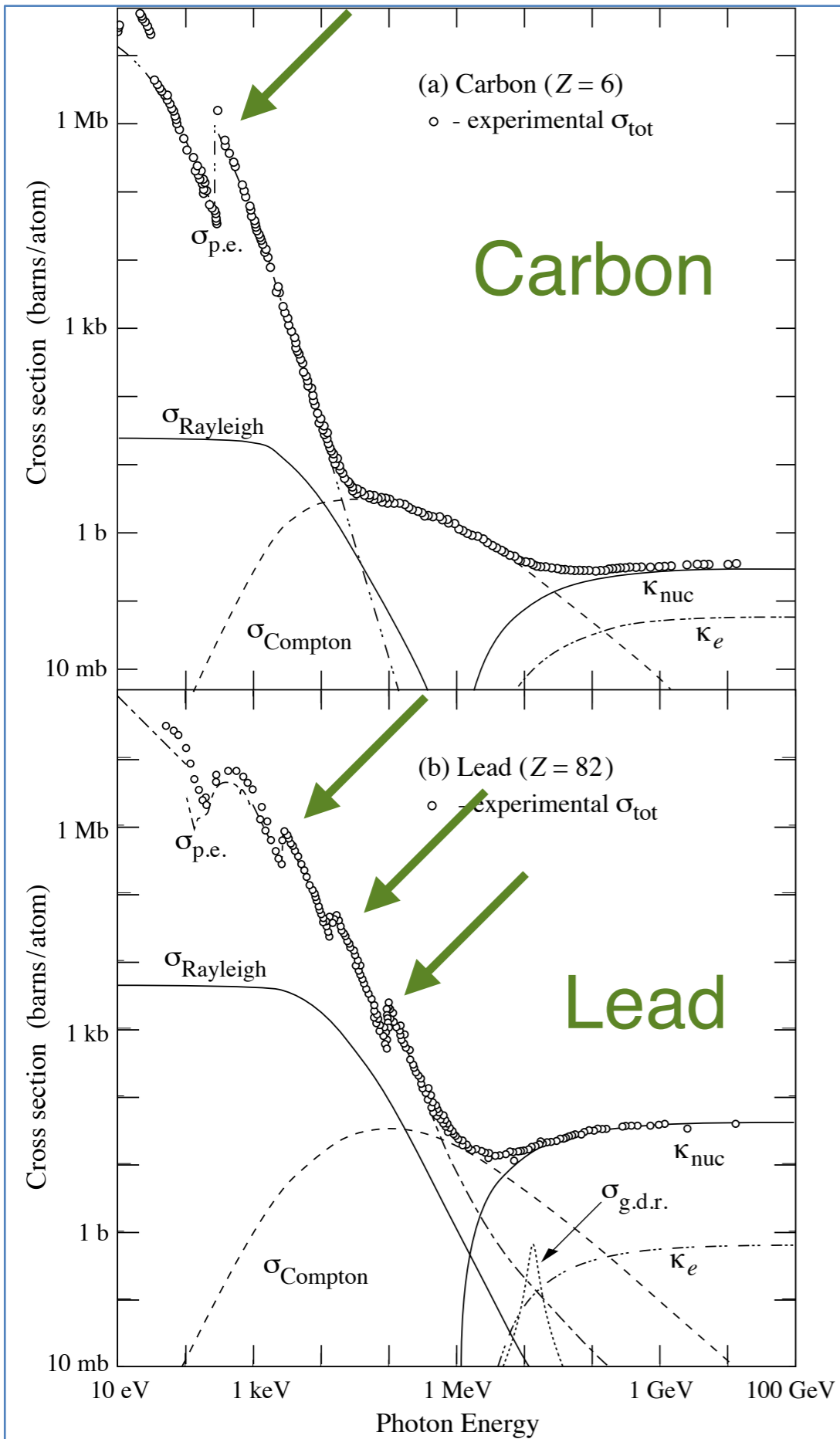
Photon energy E_γ Binding energy

Reaction is only possible if $E_\gamma > I_b$

The absorption edges correspond to the binding energies of different atomic shells -- there are extra turn-on points at which $E_\gamma > I_b$ for another shell, and so total cross-section increases.



Photons & matter: Photoelectric Effect



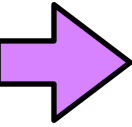
Different elements have different electron shell structures

=> The photoelectric absorption edges are at different energies.

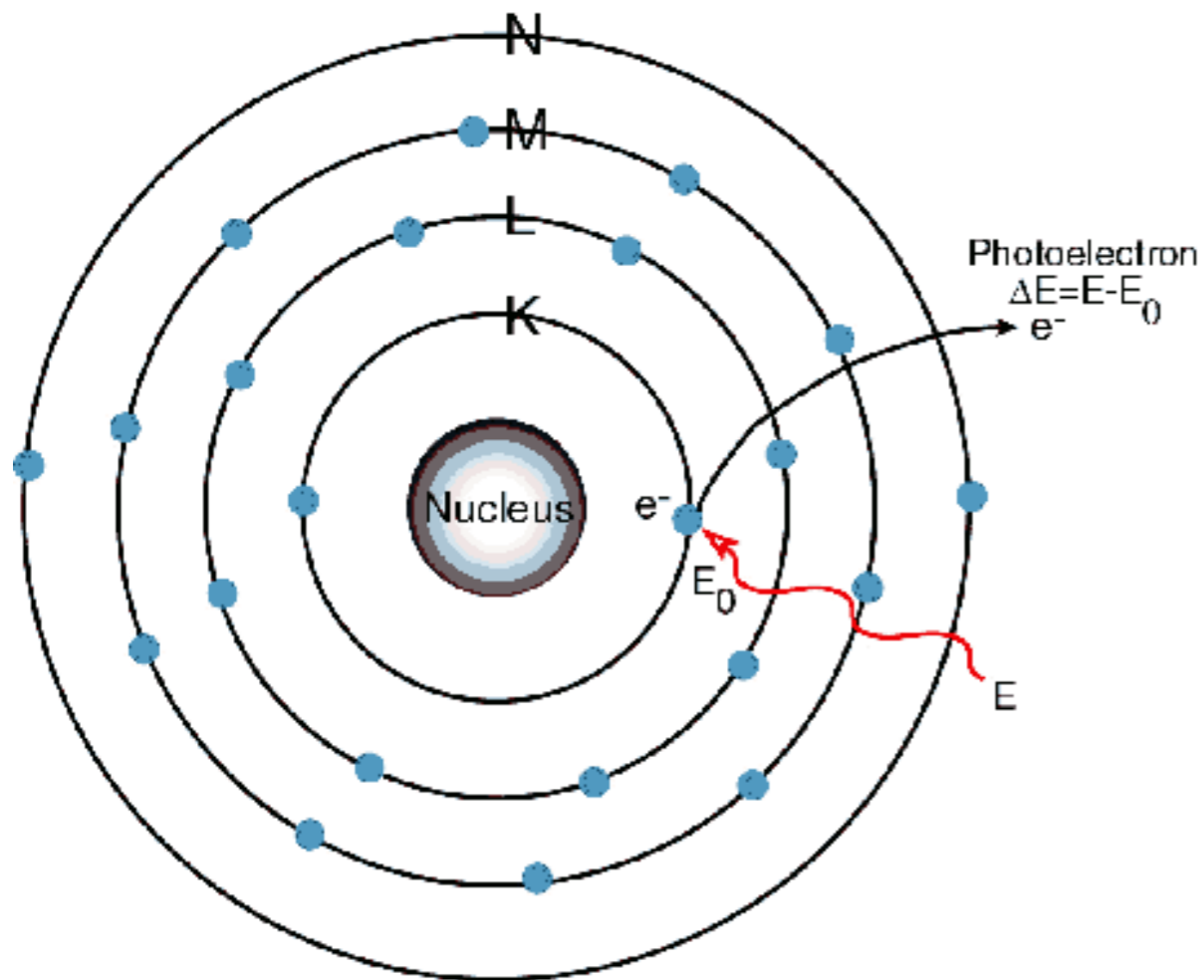
Can use these characteristic energies to identify elements...

Photons & matter: Photoelectric Effect

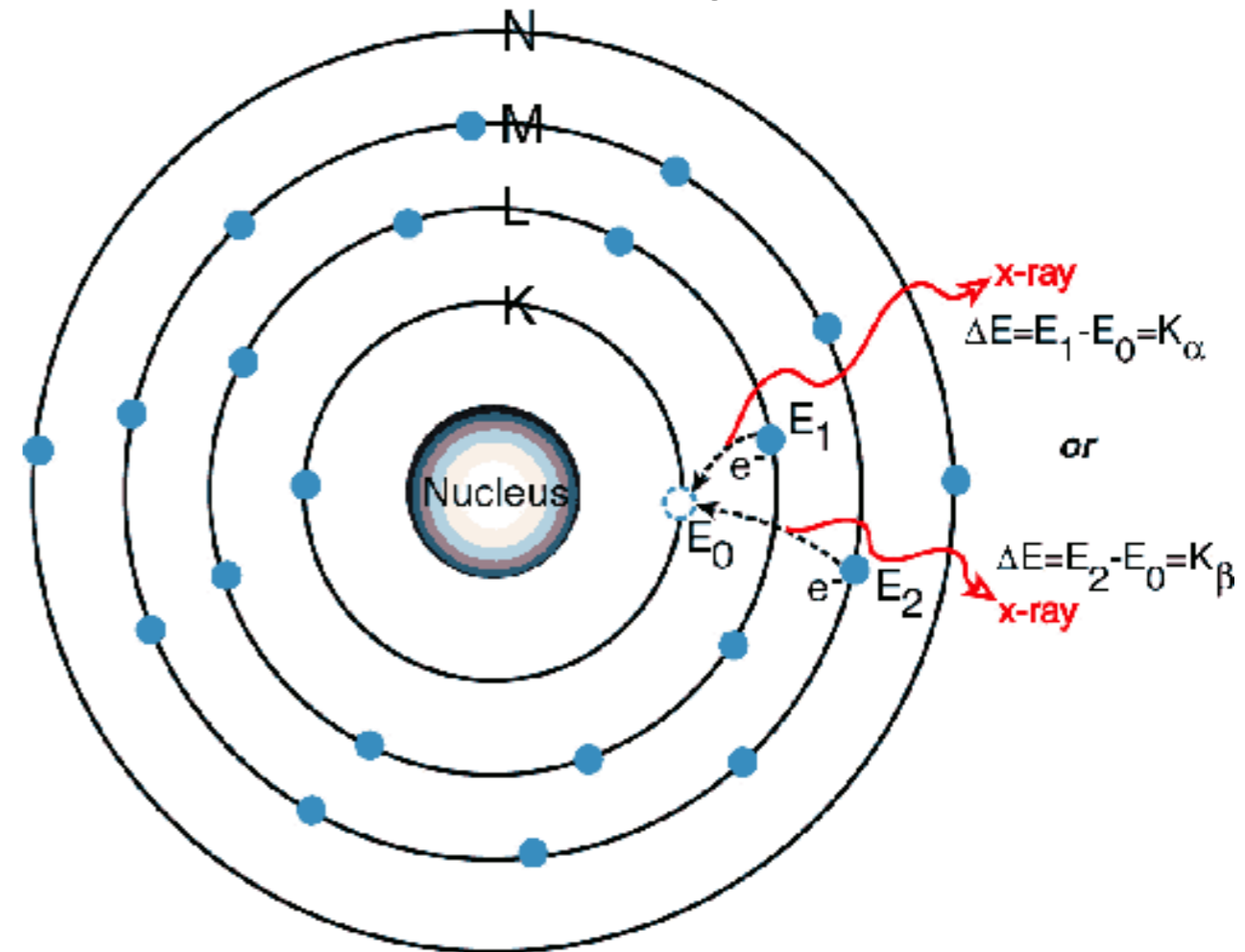
Characteristic X-ray energies



(Figures from Amptek, Inc.)



Step 1: Photoelectric effect knocks out an electron in one shell, leaving a vacancy.

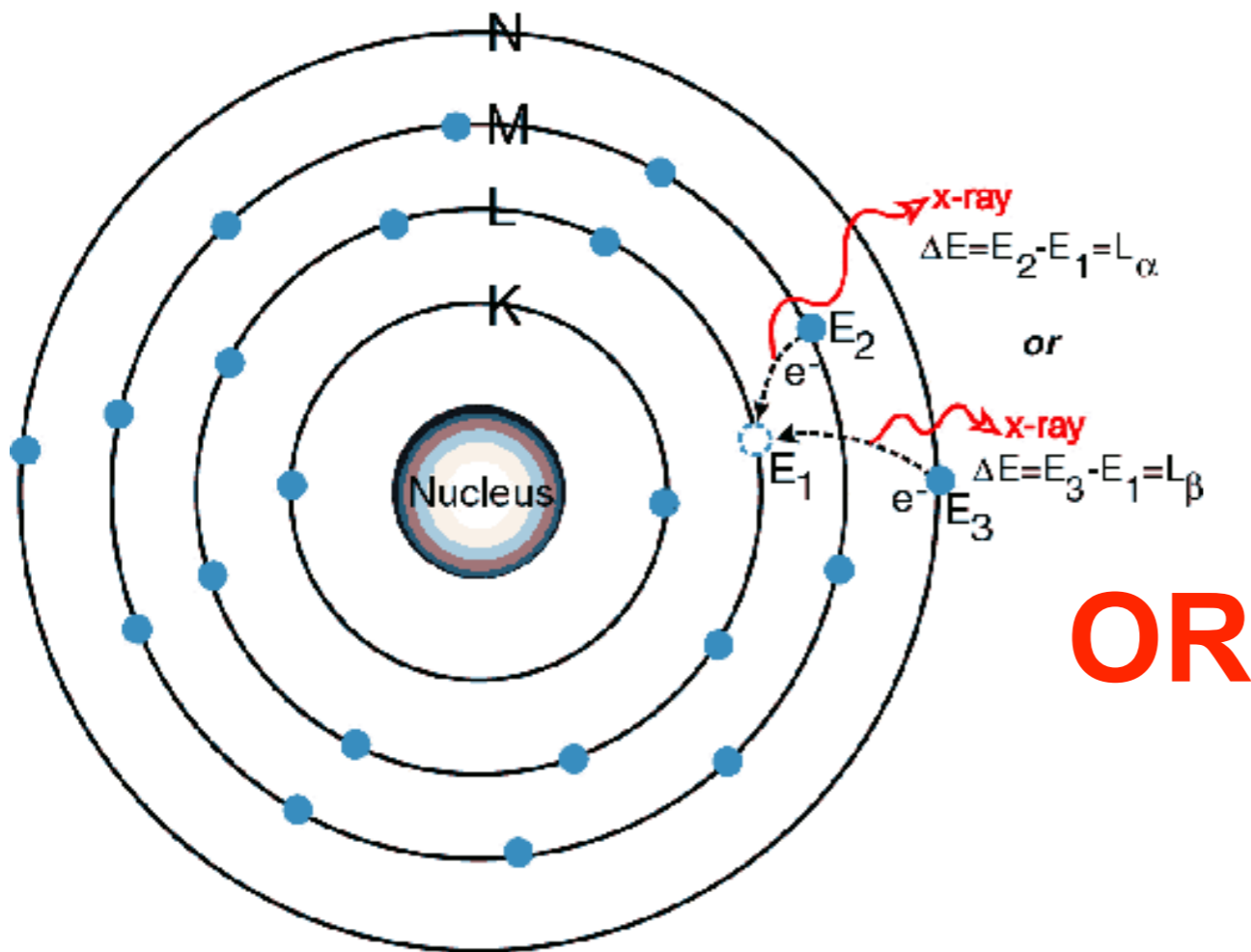


Step 2: An electron from a higher shell drops down to fill the vacancy. Potential energy is converted into an X-ray whose energy is characteristic of the element's electron shell structure. Leaves vacancy in an outer shell.

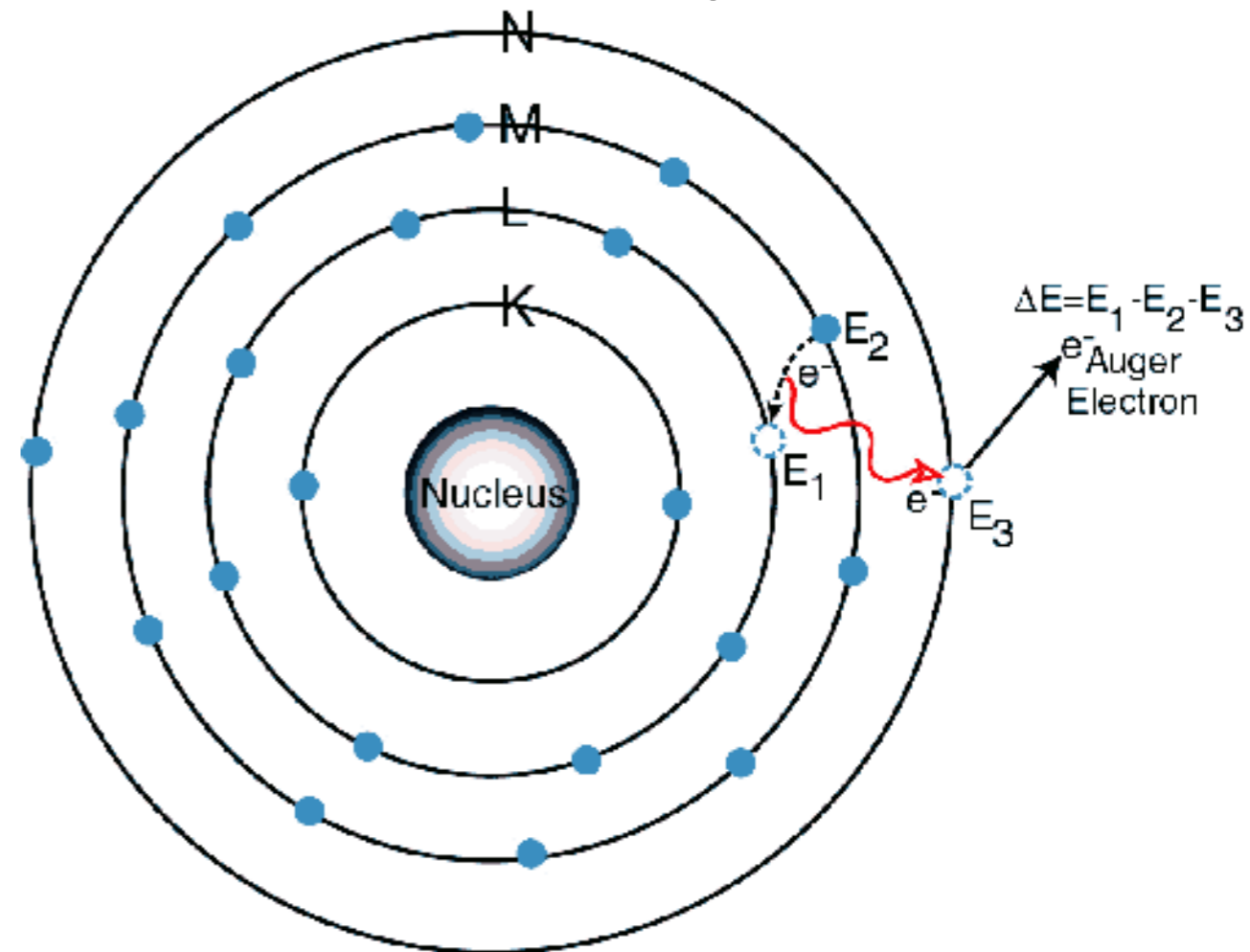
Photons & matter: Photoelectric Effect

Characteristic X-ray energies

(Figures from Amptek, Inc.)



OR



Step 3: Same process can repeat for the vacancy left in an outer shell \Rightarrow two (or more) X-rays emitted with a characteristic pattern of energies.

\Rightarrow X-ray spectroscopy

Step 3: An electron drops down, but the potential energy is instead used to liberate an electron from an outer shell (ionisation + kinetic energy). Kinetic energy of the electron is characteristic of the element.

\Rightarrow Auger spectroscopy

Photons & matter: Photoelectric Effect

Last words: note that

- The full energy of the incoming photon is absorbed by the material
- Most* of this energy ($E_\gamma - I_b$) is immediately transferred to the ejected electron
- Most of the rest (I_b) is reemitted as lower-energy X-rays and/or an Auger electron (likely to be recaptured)
- Therefore: essentially **100% of the photon's energy is seen by the detector****.

* For the energies we work with, $E_\gamma \gg I_b$

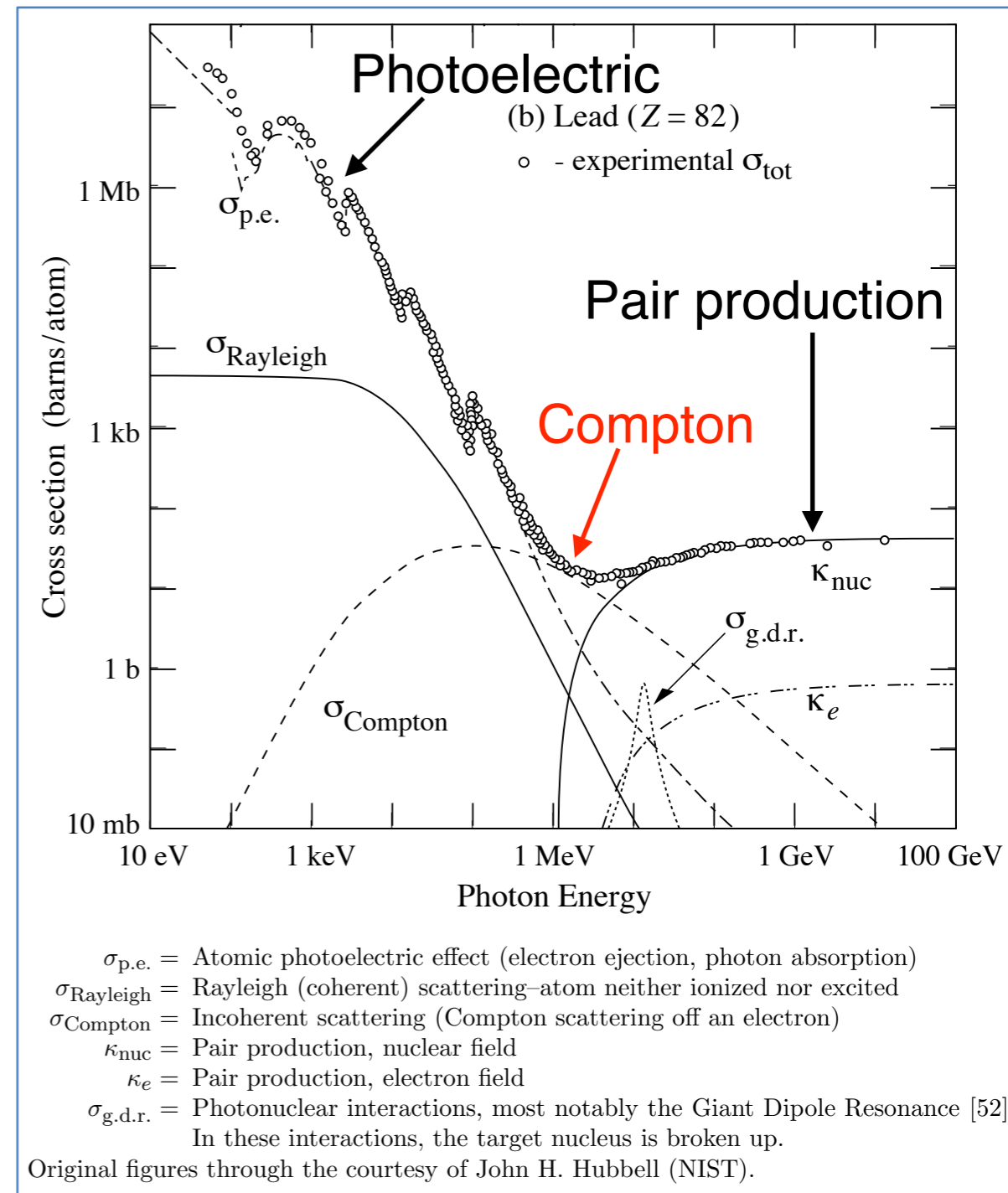
** Can fail if detector is not large compared to depth of EM shower, e.g. for very high energy photons/electrons at a collider.

Photons & matter: Compton Scattering

The dominant interaction is one of these three for photon energies above 10 eV.

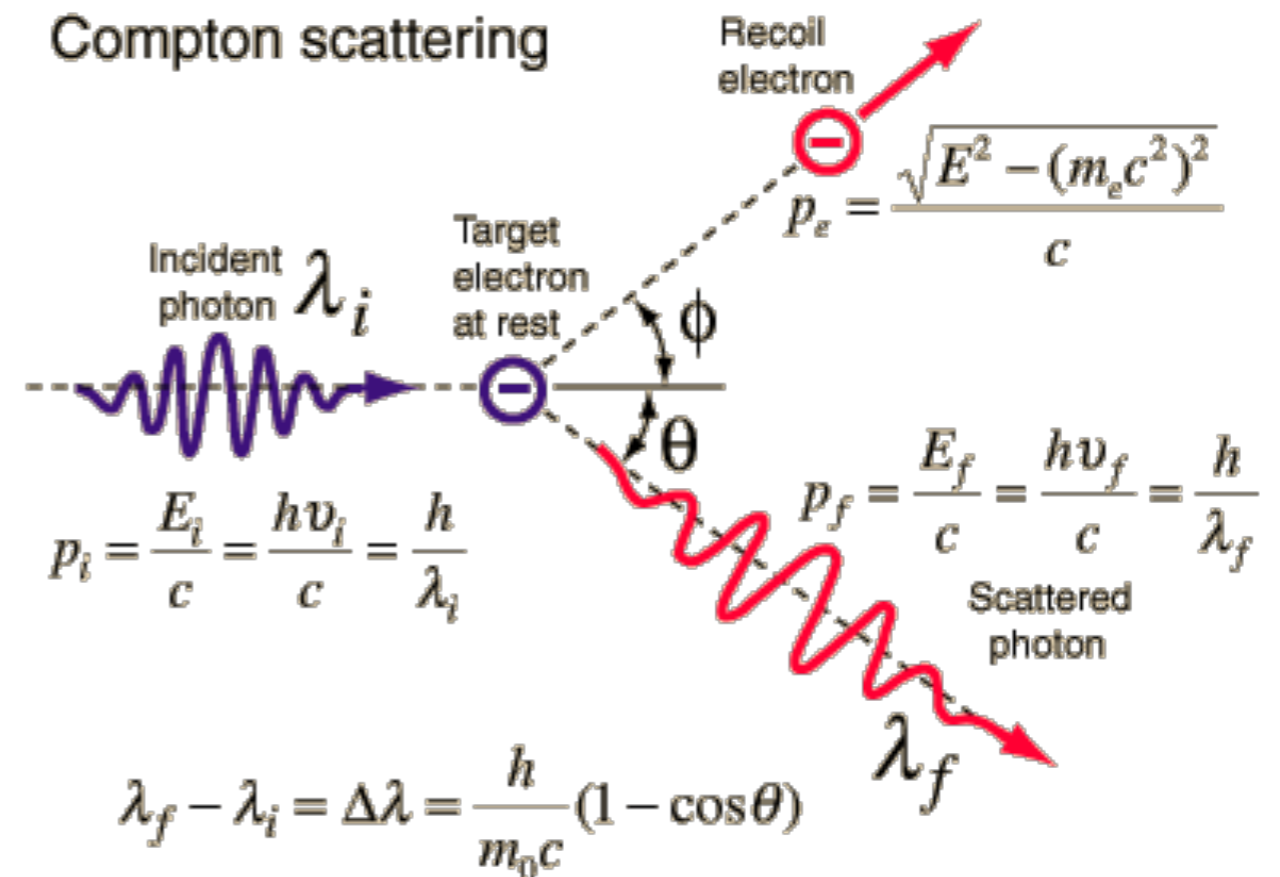
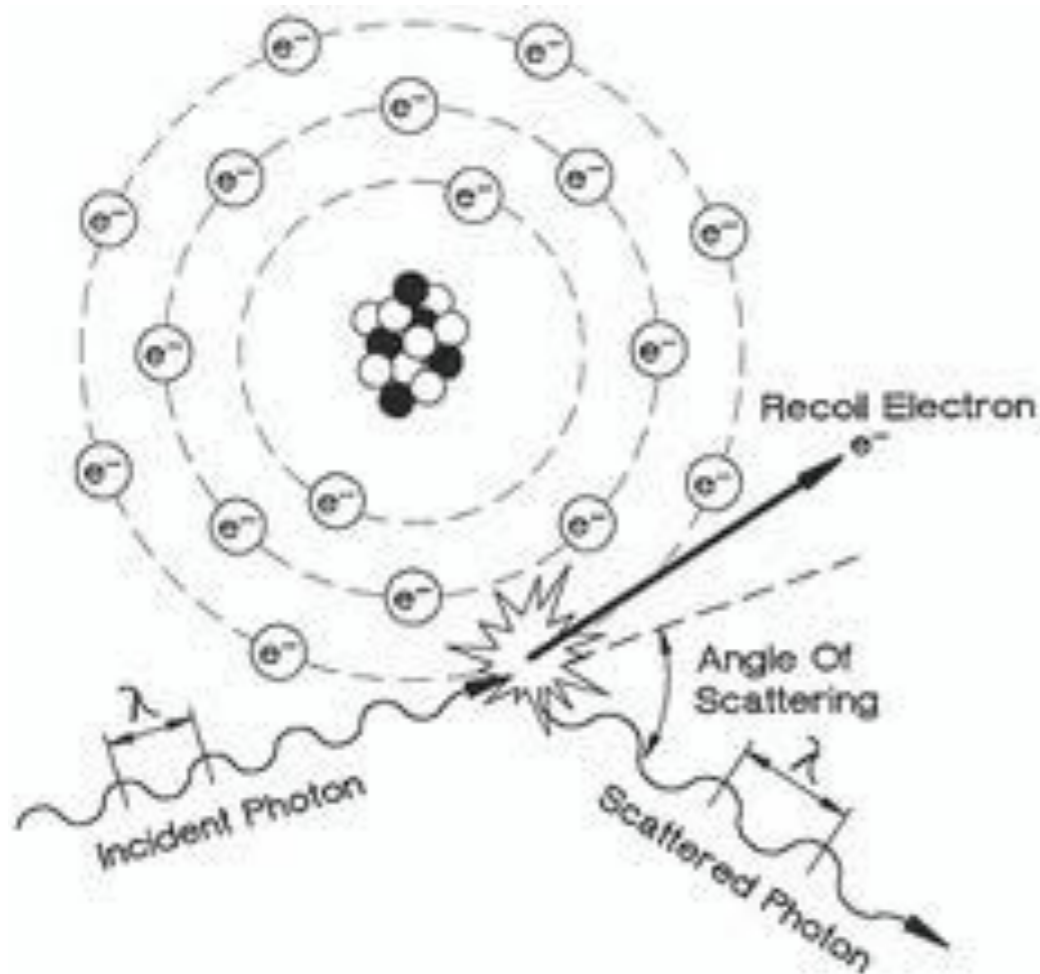
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- ...



Photons & matter: Compton Scattering

Compton scattering is **elastic scattering** of a photon on a free charged particle, usually an electron.



A bit of algebra =>

$$h\nu = \frac{h\nu_0}{1 + \alpha(1 - \cos\theta)} \quad [\text{keV}]$$

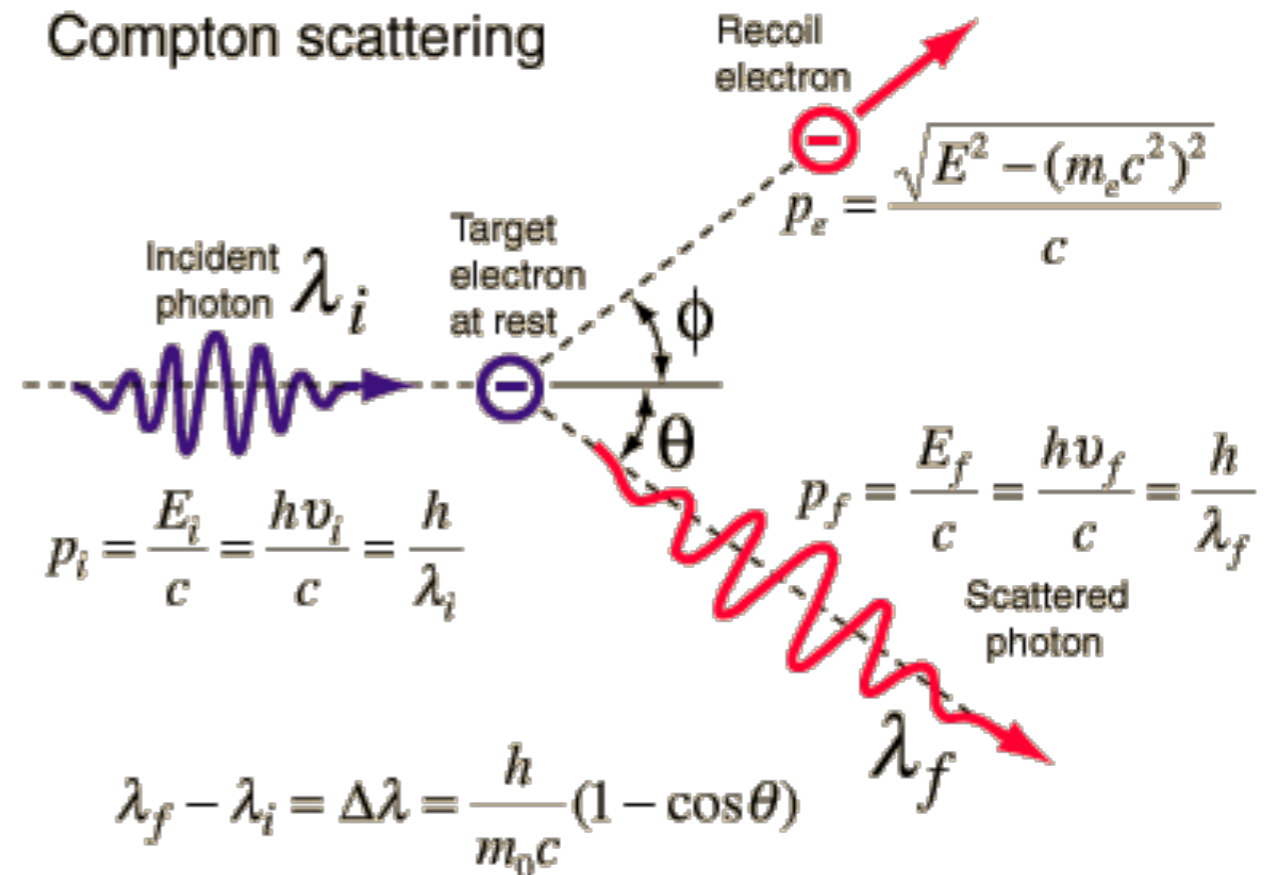
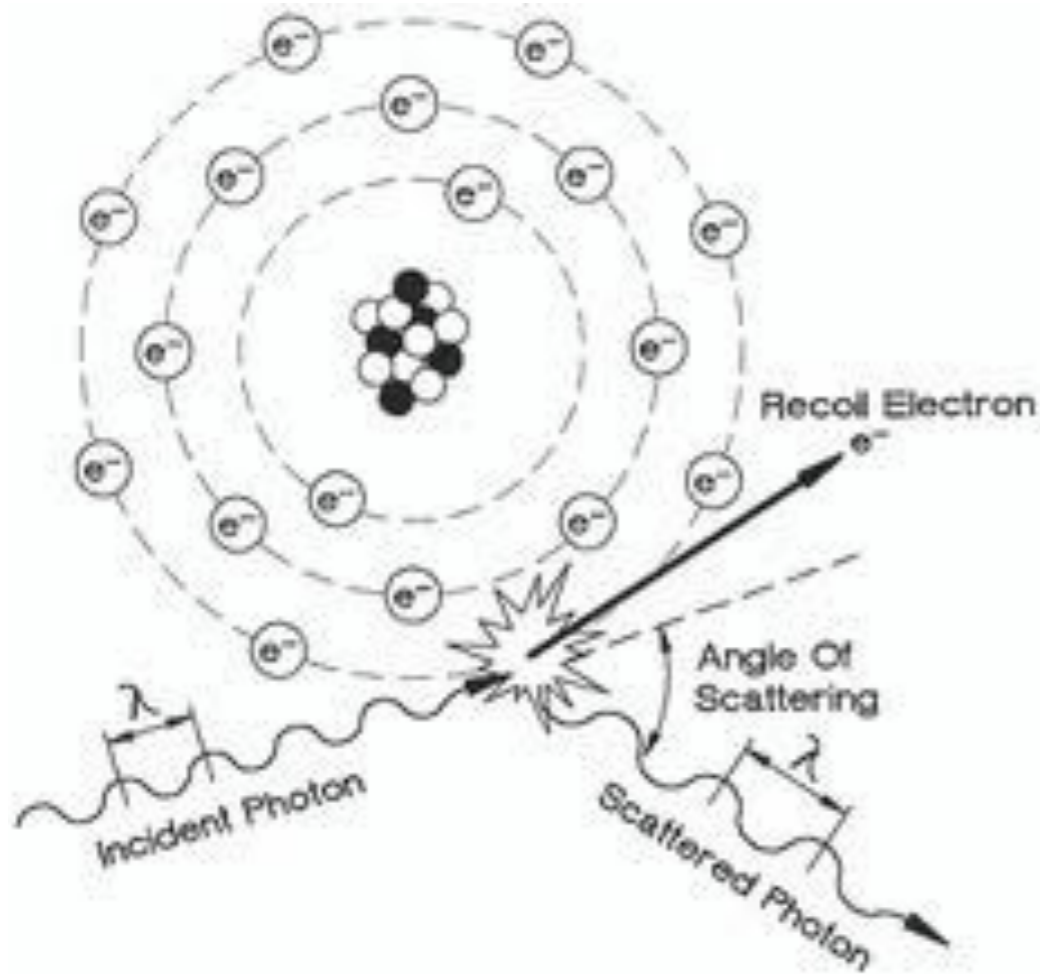
$$T_e = \frac{h\nu_0 \alpha(1 - \cos\theta)}{1 + \alpha(1 - \cos\theta)} \quad [\text{keV}]$$

where

- $h\nu_0$ = incoming photon energy
- $h\nu$ = outgoing photon energy
- $\alpha = h\nu_0 / (m_e c^2)$
- T_e = outgoing electron kinetic energy

Photons & matter: Compton Scattering

Compton scattering is quasi-elastic scattering of a photon on a quasi-free* charged particle, usually an electron. * initial photon energy \gg electron binding energy



A bit of algebra =>

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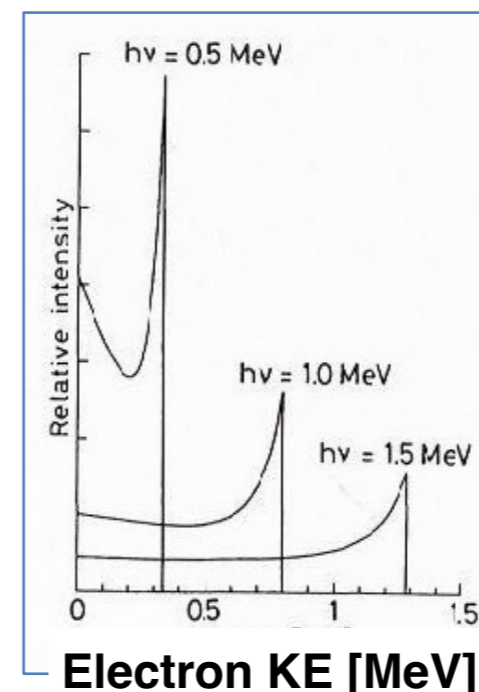
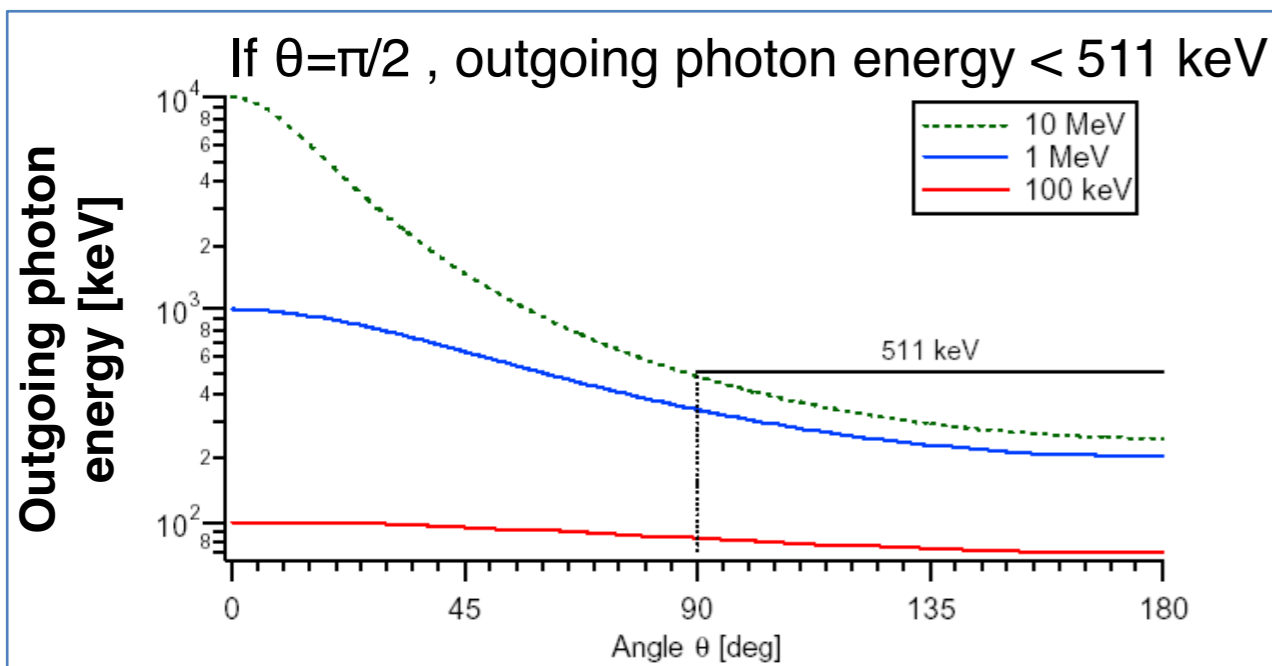
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Bounding cases:

- $\theta=0$: $h\nu = h\nu_0$; $T_e = 0 \Rightarrow$ infinitely weak scatter, **photon undeflected**
- $\theta=\pi$: $h\nu = h\nu_0 / (1 + 2\alpha)$ and $T_e = T_{e,\text{max}} = 2\alpha h\nu_0 / (1 + 2\alpha)$
 \Rightarrow **maximum recoil energy (Compton edge)**



Note: photon always exits with some energy ($T_e < h\nu_0$)
 \Rightarrow If photon doesn't interact again, the detector won't see the remaining energy ($h\nu$).

Photons & matter: Compton Scattering

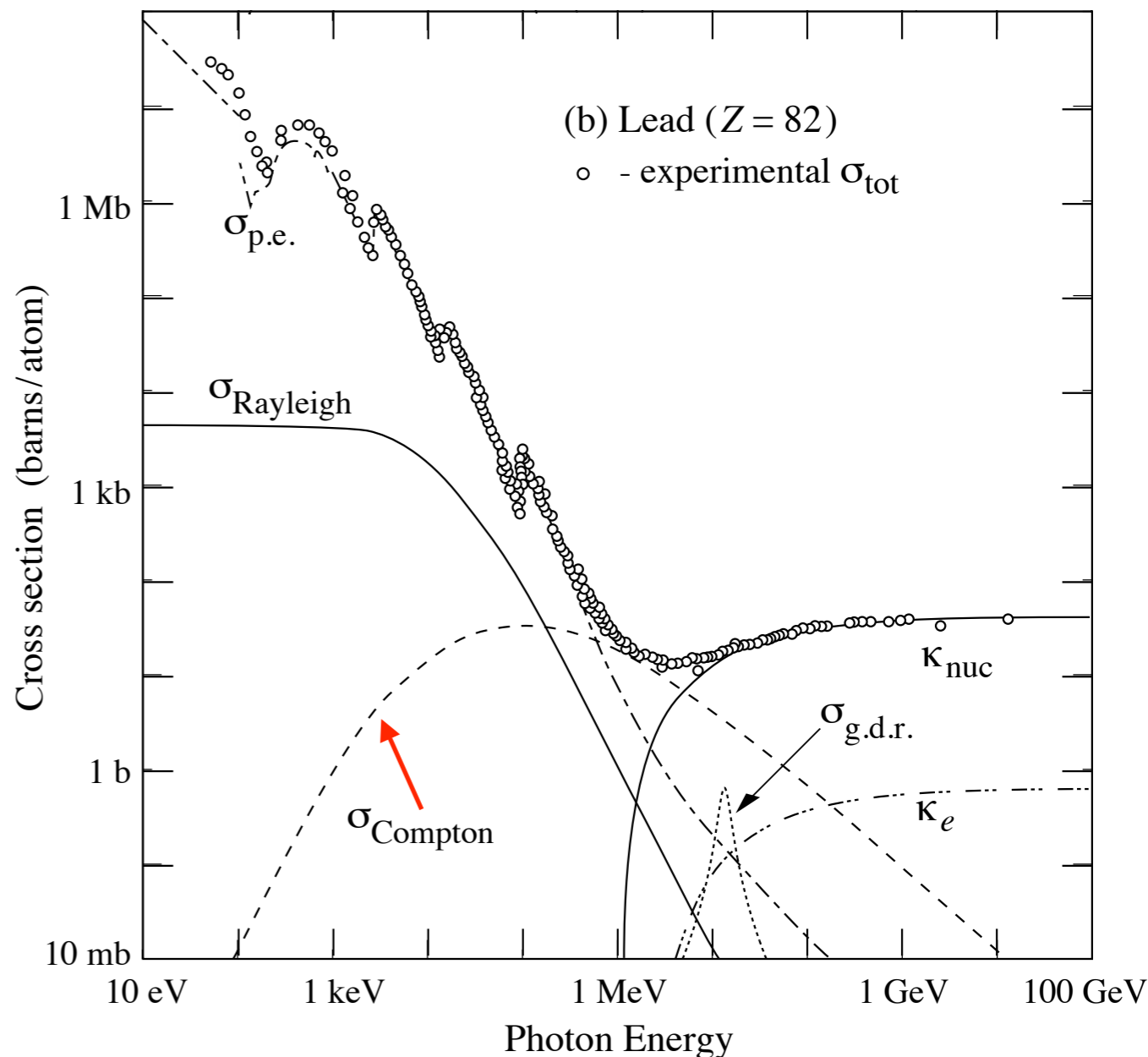
Cross-section calculated by Klein & Nishina (1929):

$$\frac{d\sigma_{\text{KN}}(h\nu_0, \theta)}{d\Omega} = \frac{r_e^2}{2} \left(\frac{1 + \cos^2 \theta}{(1 + \alpha(1 - \cos \theta))^2} + \frac{\alpha^2 (1 - \cos \theta)^2}{(1 + \alpha(1 - \cos \theta))^3} \right)$$

where r_e is the classical electron radius.

(Formula assumes initial electron is free; corrections required if incoming photon energy is low or Z is high.)

Important from (10-100 keV) up through (10-100 MeV), depending on the material.



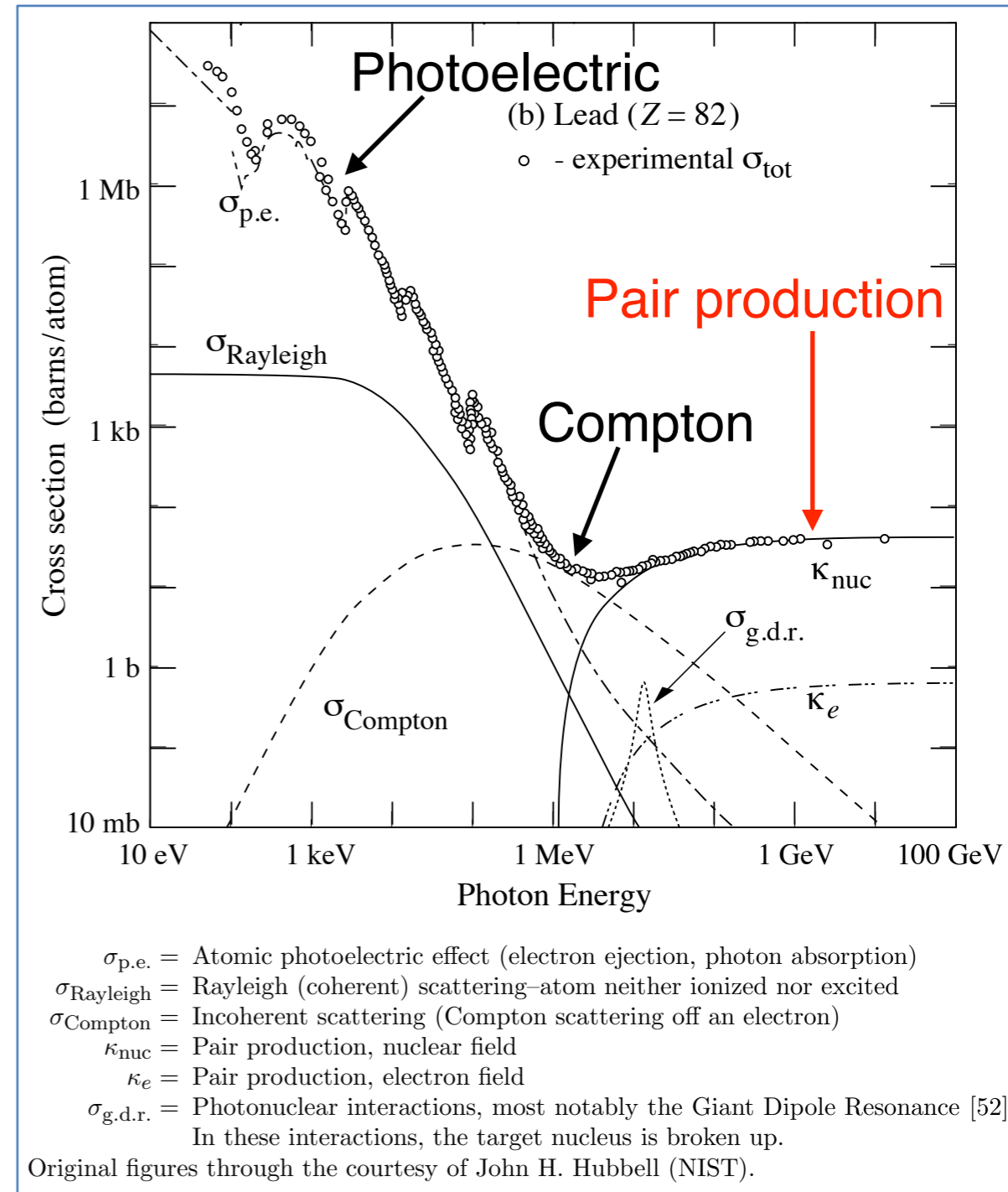
$$r_e = e^2 / 4\pi\epsilon_0 m_e c^2 \approx 2.82 \text{ fm}$$

Photons & matter: Pair production

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→ Photons & matter: Pair production

Energy threshold for pair production:

$$E_\gamma \geq 2m_e c^2 \left(1 + \frac{m_e}{m_N} \right)$$

Cross-section ramps up from threshold, then saturates.

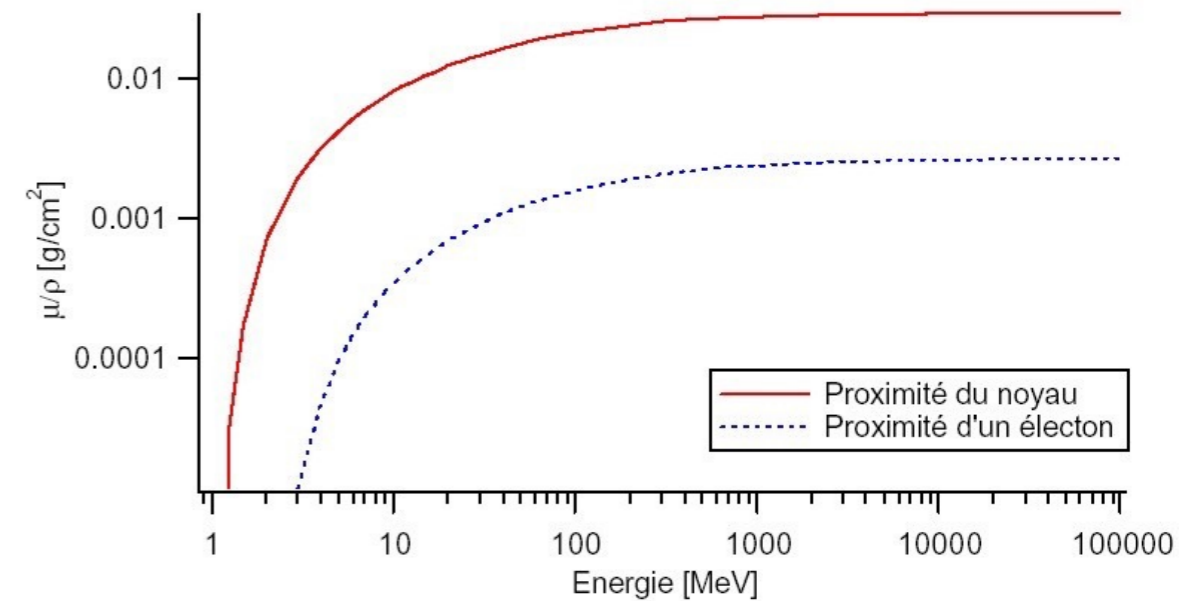
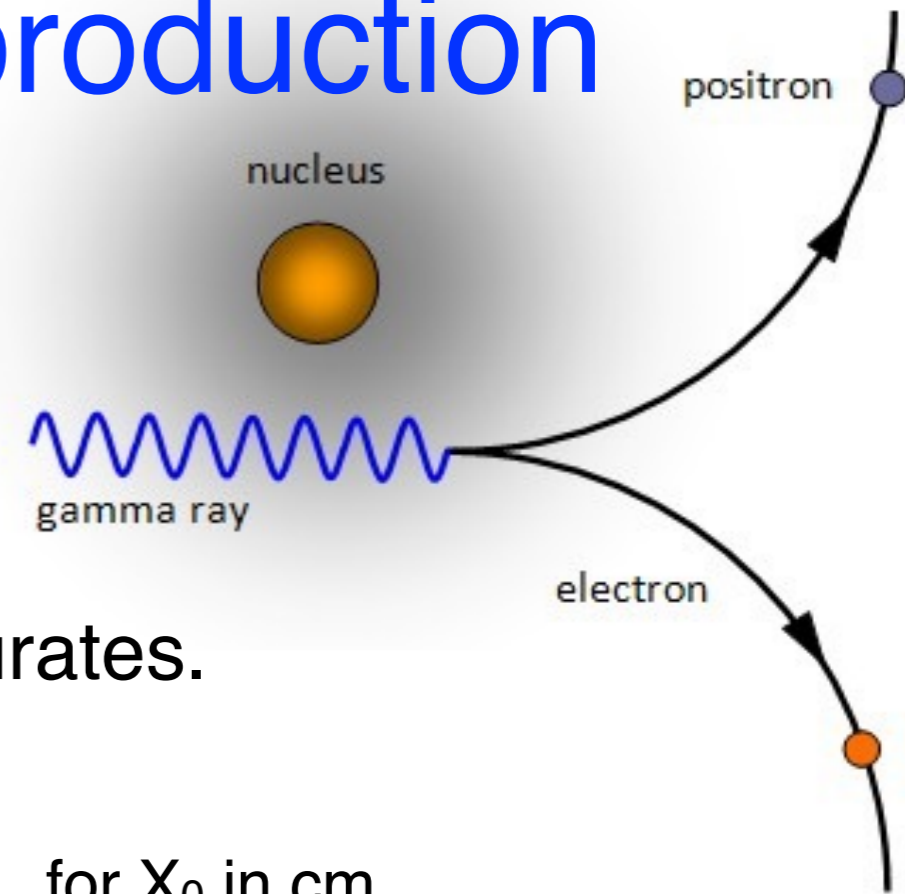
For $E_\gamma \geq 2m_e c^2$,

$$\sigma_{\text{pair}} = 4\alpha r_e^2 Z^2 \left(\frac{7}{9} \ln \frac{183}{Z^{1/3}} - \frac{1}{54} \right) \approx \frac{7}{9} \frac{A}{N_A} \frac{1}{\rho X_0} \text{ cm}^2 \quad \text{for } X_0 \text{ in cm [caution!]}$$

Recall: mean free path $\lambda = 1/\mu = 1/(n\sigma)$
and $n = \rho N_A / A$ for atoms/nuclei, thus

$$\lambda_{\text{pair}} = \frac{1}{n\sigma} = \frac{1}{\frac{\rho N_A}{A} \frac{7}{9} \frac{A}{N_A} \frac{1}{\rho X_0}} = \frac{9}{7} X_0$$

... so **pair production mean free path $\sim X_0$**
for $E_\gamma \geq 2m_e c^2$



Q: Why is the reaction $\gamma \rightarrow e^+ e^-$ impossible? What is the role of the nucleus in pair production?

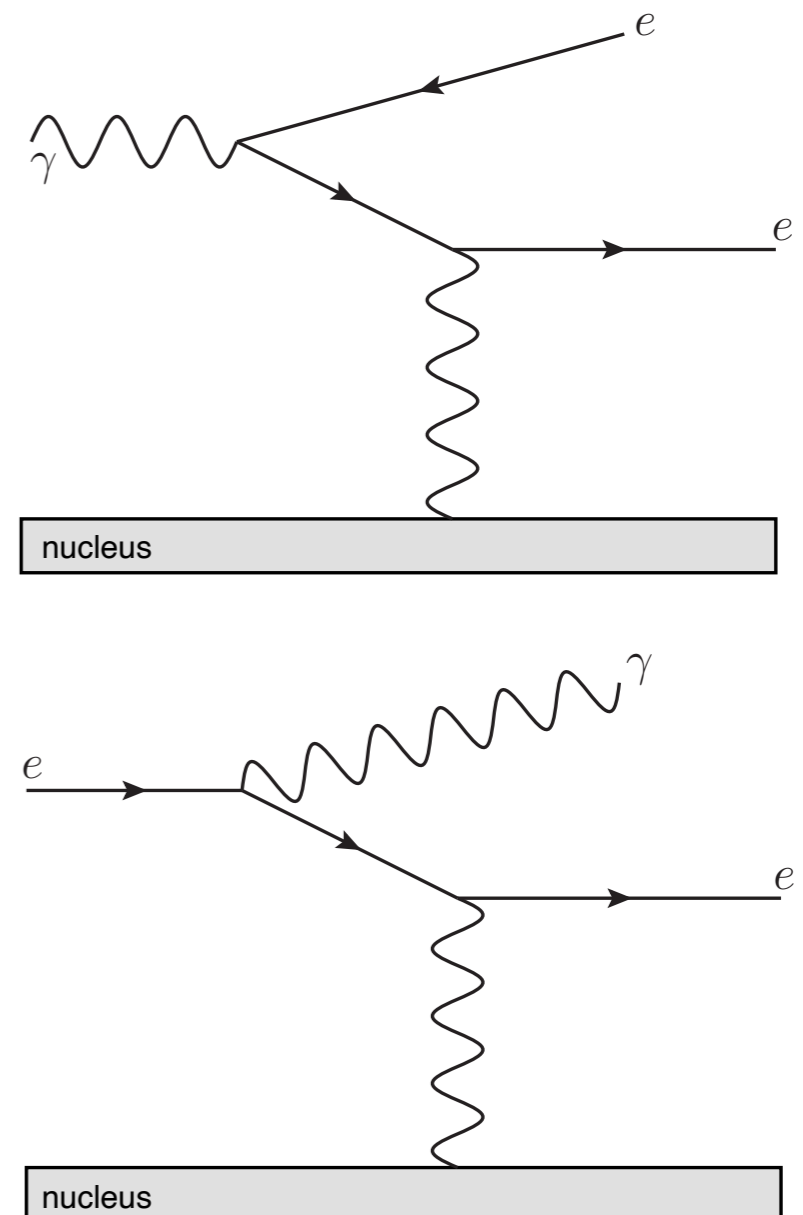
Photons & matter: Pair production

Remember that radiation length X_0 was defined as the distance over which the energy of an electron/positron is reduced by a factor of $1/e$ through radiation losses.

Fundamental connection between the two processes.

$$\lambda_{\text{pair}} = \frac{1}{n\sigma} = \frac{1}{\frac{\rho N_A}{A} \frac{7}{9} \frac{A}{N_A} \frac{1}{\rho X_0}} = \frac{9}{7} X_0$$

... so pair production mean free path $\sim X_0$
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Photons & matter: Pair production

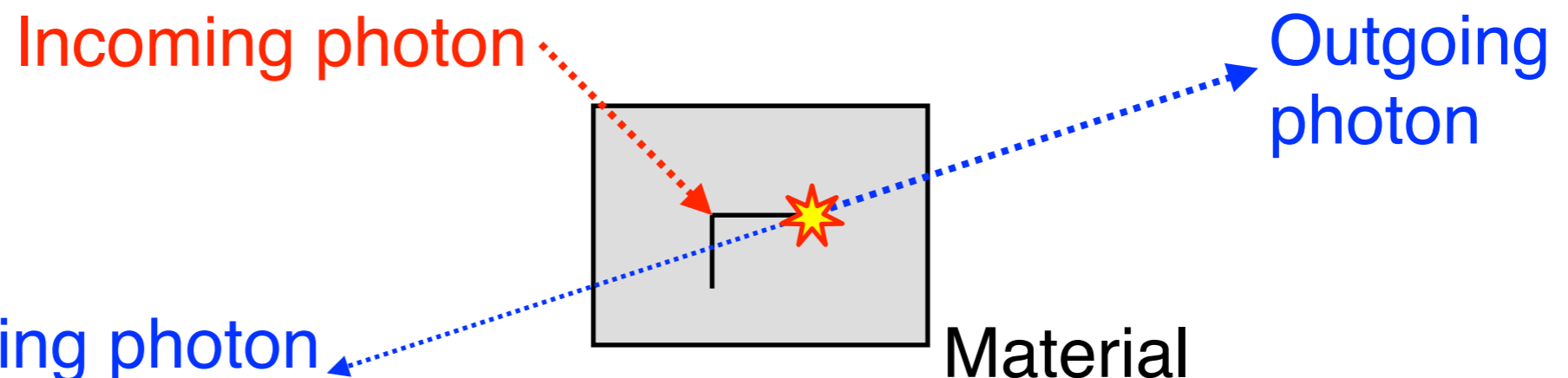
What happens to **positrons** emitted in pair production?

- They **lose energy** in matter (much like electrons)
- They **annihilate** with an electron
- Former is a stochastic process spread out over time; latter is all-or-nothing

In practice:

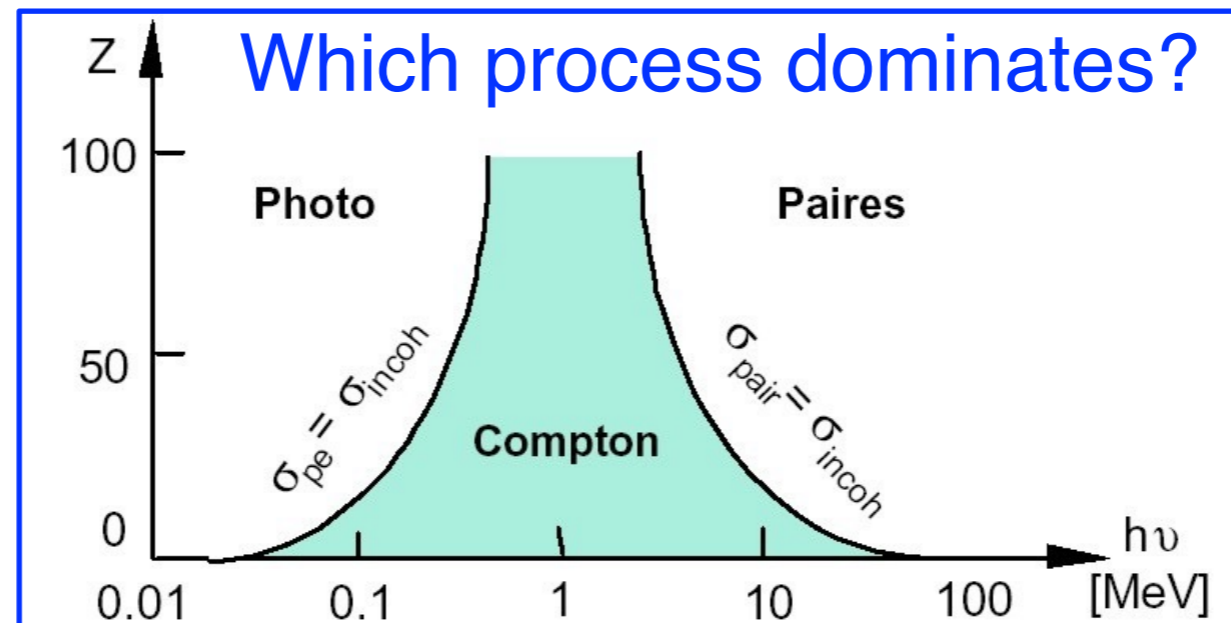
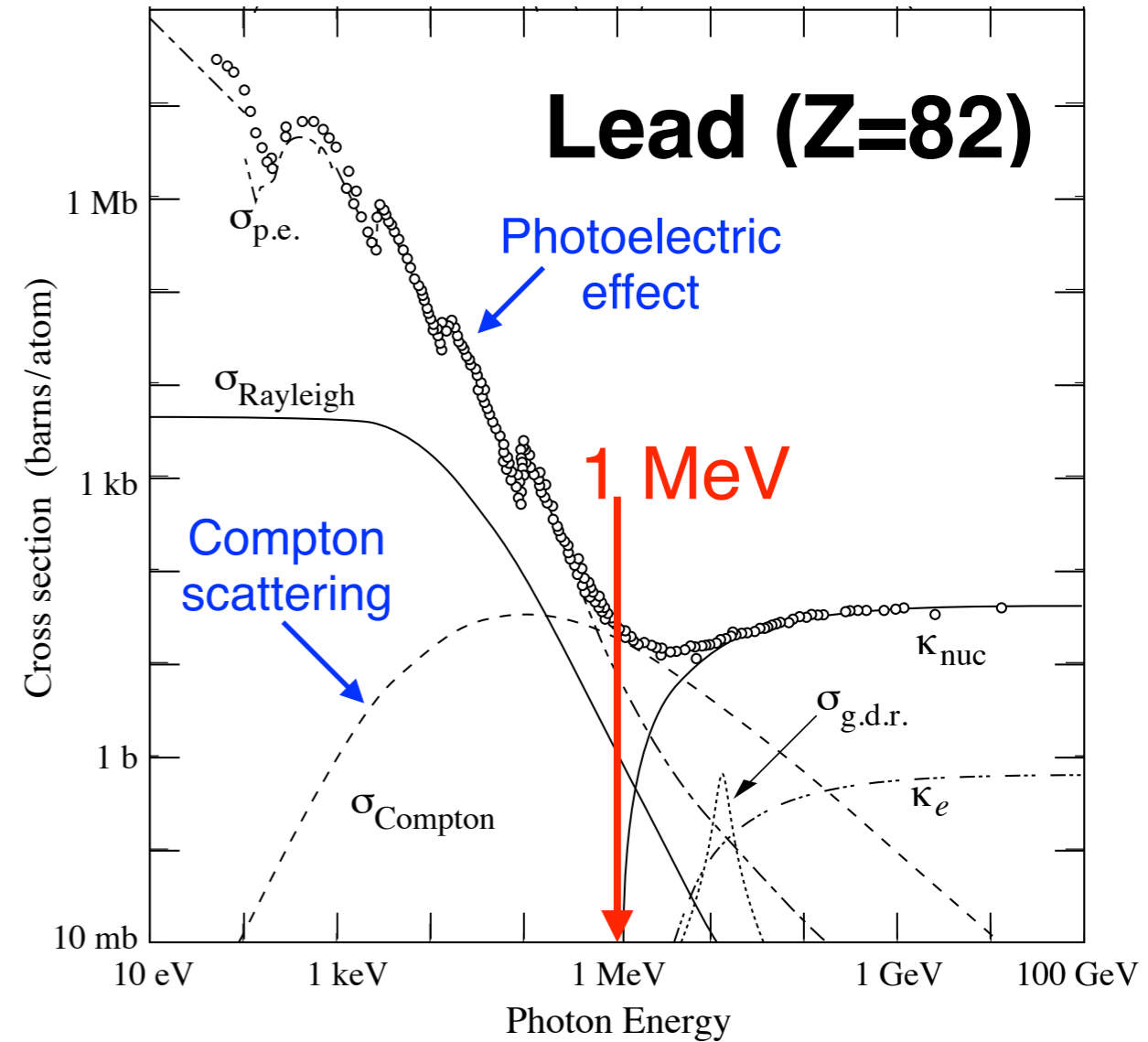
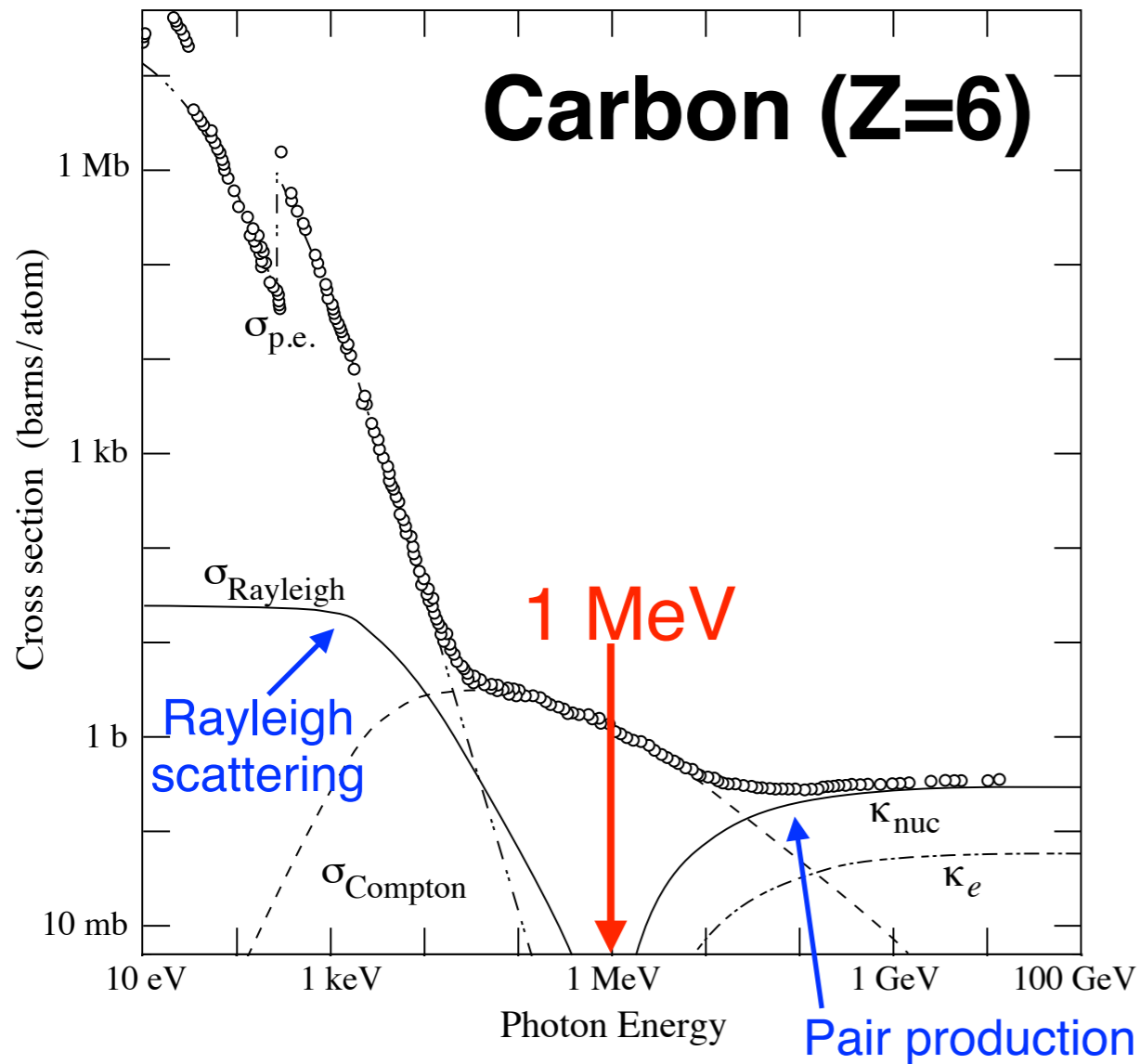
- The energy loss happens more quickly => positrons mostly stopped before annihilating with an electron
 - Stopping time \sim few ps in a solid state detector.
- Therefore, usually assume that the **positrons annihilate at rest** with a free electron and emit two photons (or more rarely three).
- Positron can also form bound a state of e^+e^- (positronium).

As the e^+ and free electron are at rest*, or the positronium is at rest, **the two photons are emitted back-to-back***

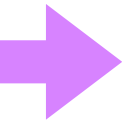


* almost

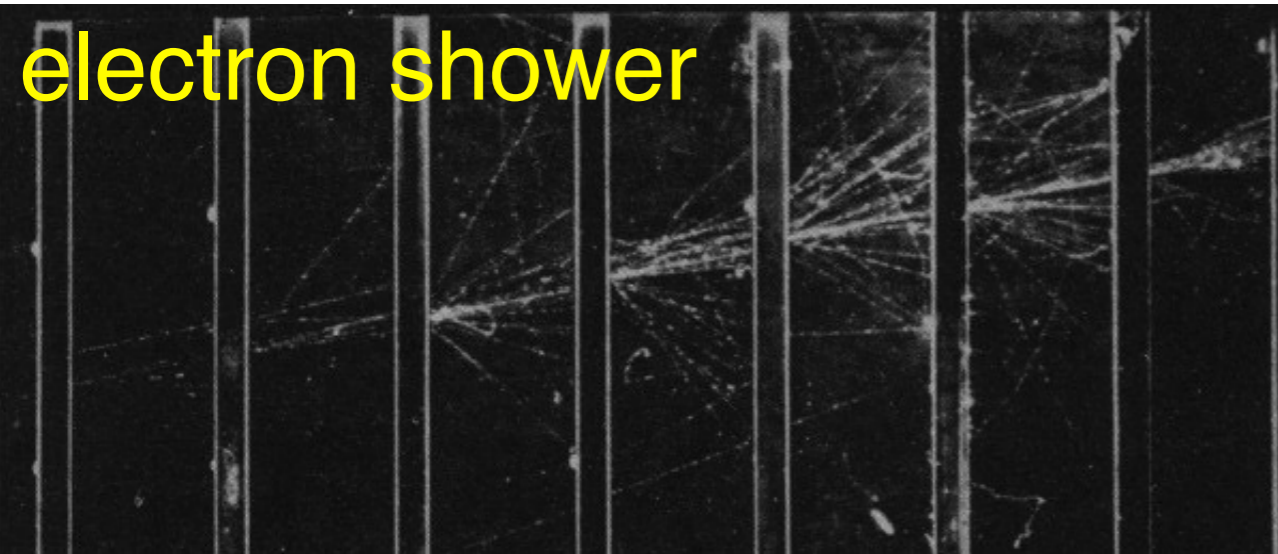
Photons & matter: Total photon x-sec



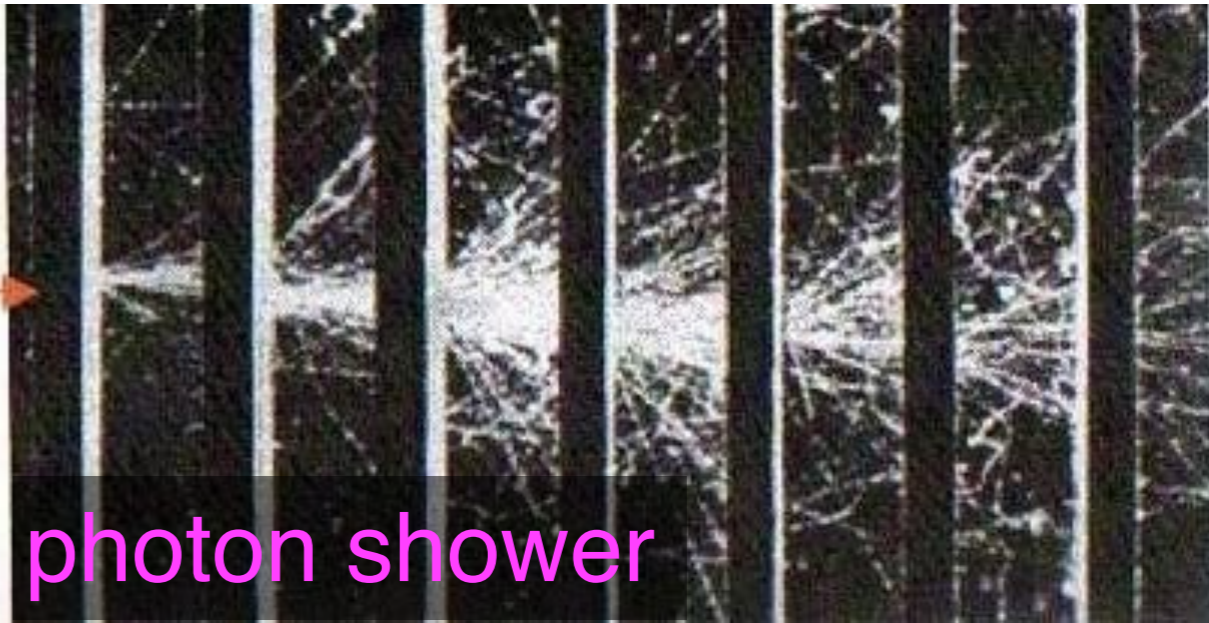
Electromagnetic showers



electron shower



Electron shower in a cloud chamber with lead absorbers



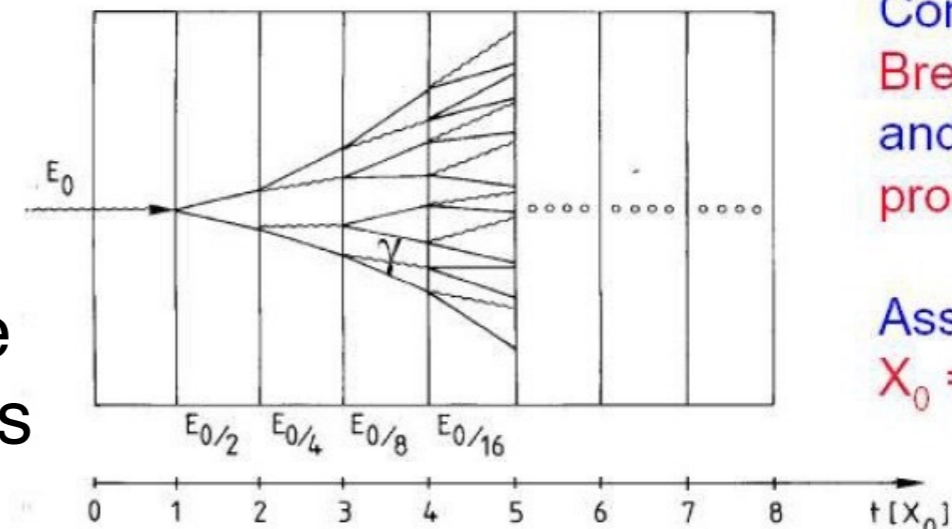
photon shower

For high energy γ and e^\pm ($E \gg m_e c^2$), showers look very similar:

- An interaction happens (on average) once per radiation length, predominantly:
 - Bremsstrahlung ($e^\pm \rightarrow e^\pm \gamma$) or pair production ($\gamma \rightarrow e^+ e^-$) [quasi-spectator nucleon omitted]
- Both reactions are $1 \rightarrow 2$ for the EM shower particles ($\gamma/e^+/e^-$)
- So after t radiation lengths, $N(t) \sim 2^t$ particles, each of avg energy $E(t) \sim E_0/2^t$
- Shower stops at $t=t_{\max}$, when $E(t)$ falls below critical energy E_c needed to sustain it

- $E(t_{\max}) = E_0 / 2^{t_{\max}} = E_c$
 $\Rightarrow t_{\max} = \ln(E_0/E_c) / \ln(2) \propto \ln(E_0)$

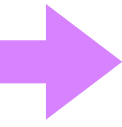
- $N(t_{\max}) = 2^{t_{\max}} = E_0/E_c \propto E_0$ -- and will be an \sim equal mix of e^+ , e^- , $\gamma \Rightarrow 1/3$ photons



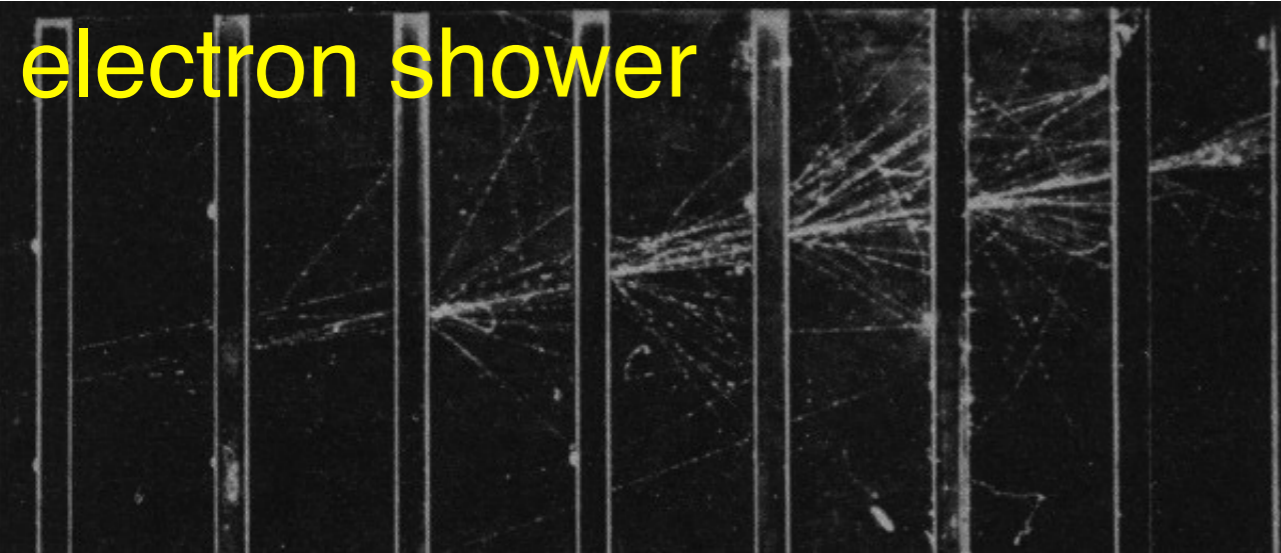
Consider only
Bremsstrahlung
and pair
production.

Assume:
 $X_0 = \lambda_{\text{pair}}$

Electromagnetic showers



electron shower



Electron shower in a cloud chamber with lead absorbers

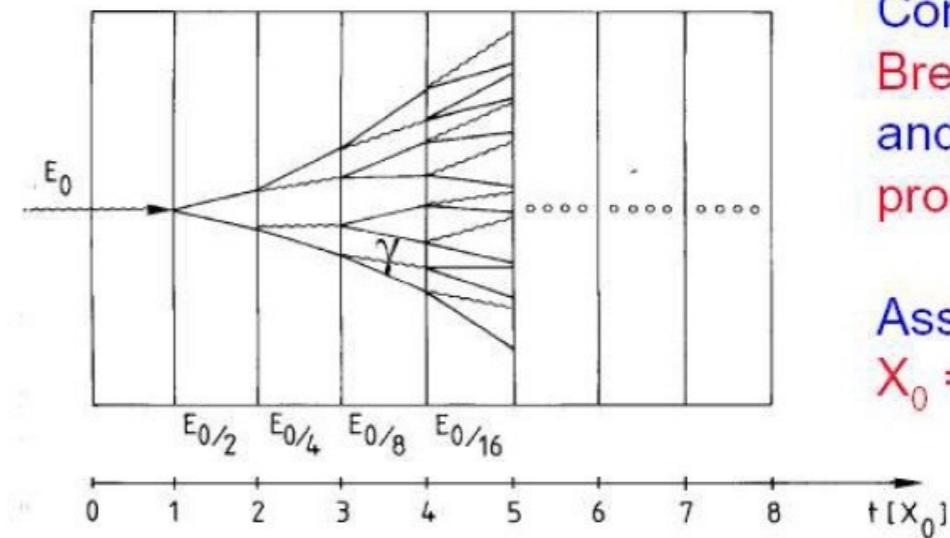


photon shower

For high energy γ and e^\pm ($E \gg m_e c^2$), showers look very similar:

- Transverse development:
- 90% of particles stay within a cylinder of radius R_M (Molière radius) around shower axis.
- $R_M = X_0 E_s / E_c$ with $E_s = 21 \text{ MeV}$

To distinguish photons from electrons/positrons, use a tracker or a thin preshower detector (thickness $\ll X_0$)

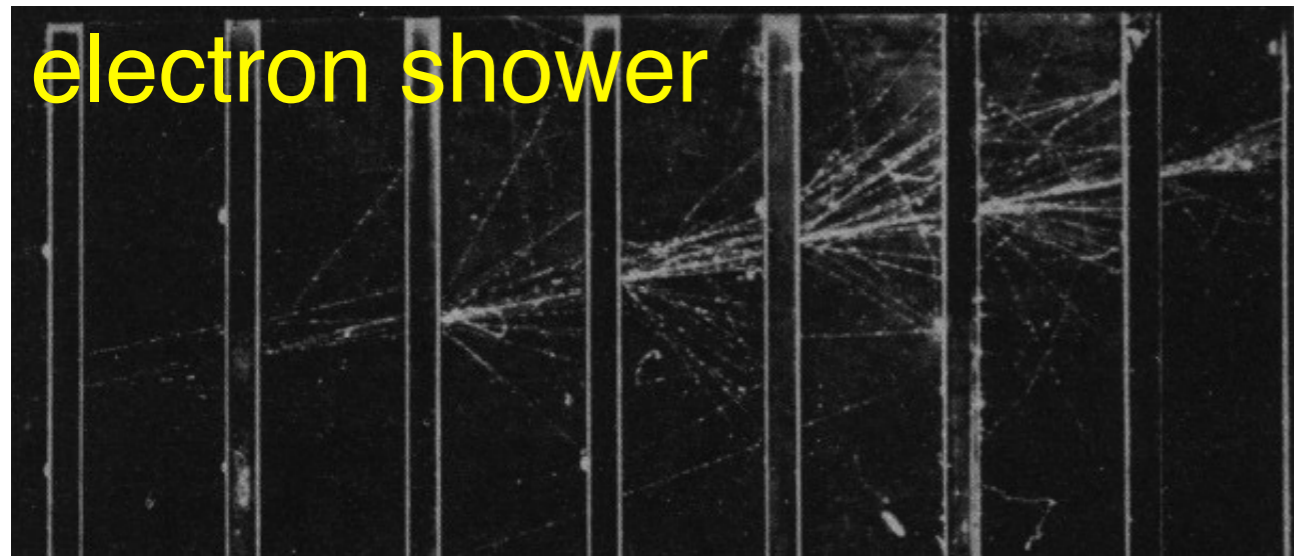


Consider only
Bremsstrahlung
and pair
production.

Assume:
 $X_0 = \lambda_{\text{pair}}$

Electromagnetic showers

electron shower



photon shower

Electron shower in a cloud chamber with lead absorbers

We said:

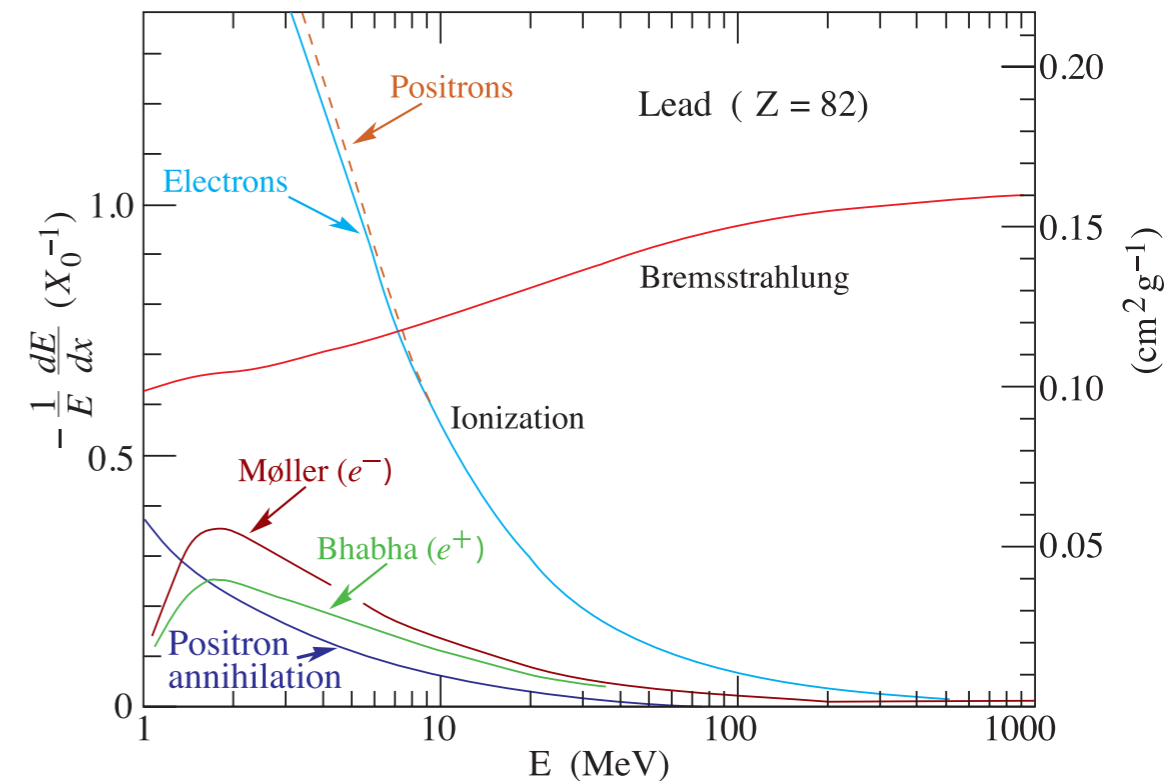
- Shower stops at $t=t_{\max}$, when $E(t)$ falls below critical energy E_c needed to sustain it

What is the critical energy?

Few tens of MeV. For energy loss of e^\pm :

- Below E_c , dominated by collision/ionisation
- Above E_c , dominated by bremsstrahlung

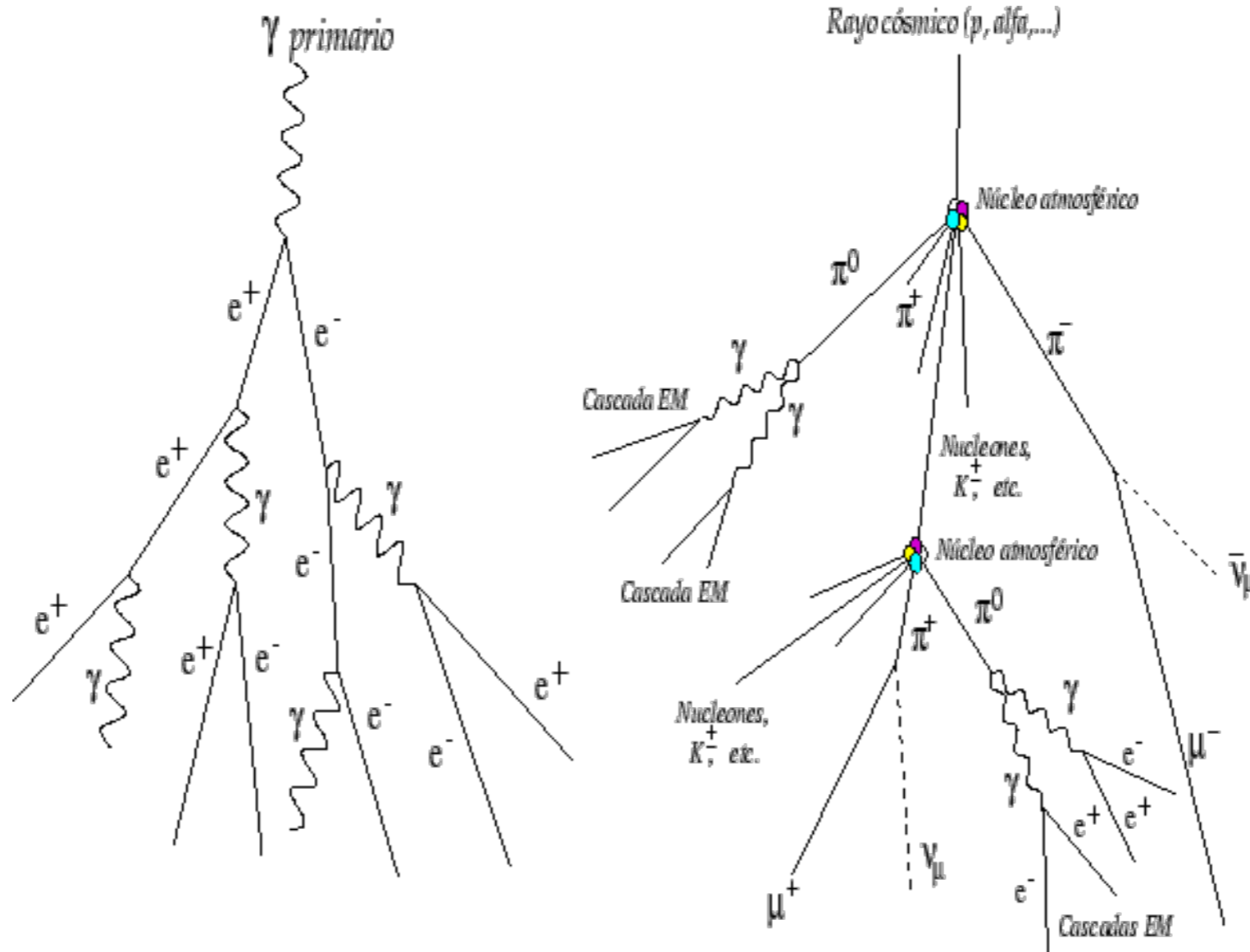
(For photons: changeover from Compton scattering to pair production occurs around few MeV too.)



Particle showers

High-energy hadrons (and...) can also shower.

Will hear more about particle showers in the calorimetry lectures.



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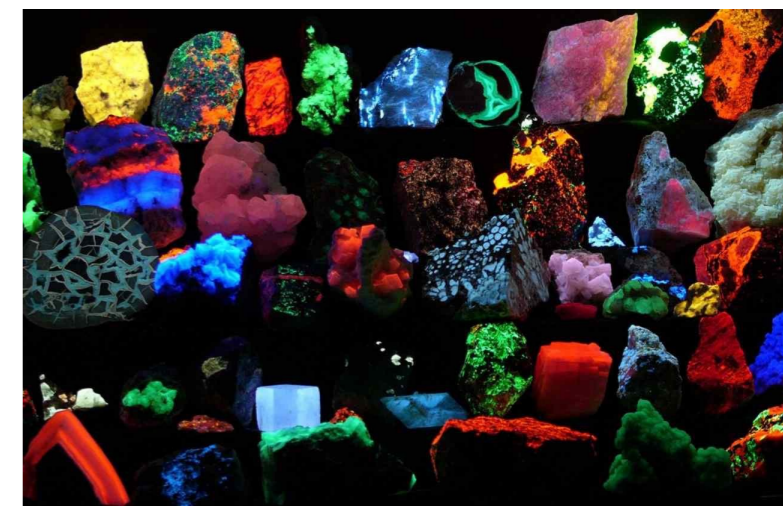
*Noyaux
Particules
Astroparticules
Cosmologie*

Master 2 Recherche

Bruno Mazoyer - LAT Orsay

Scintillators

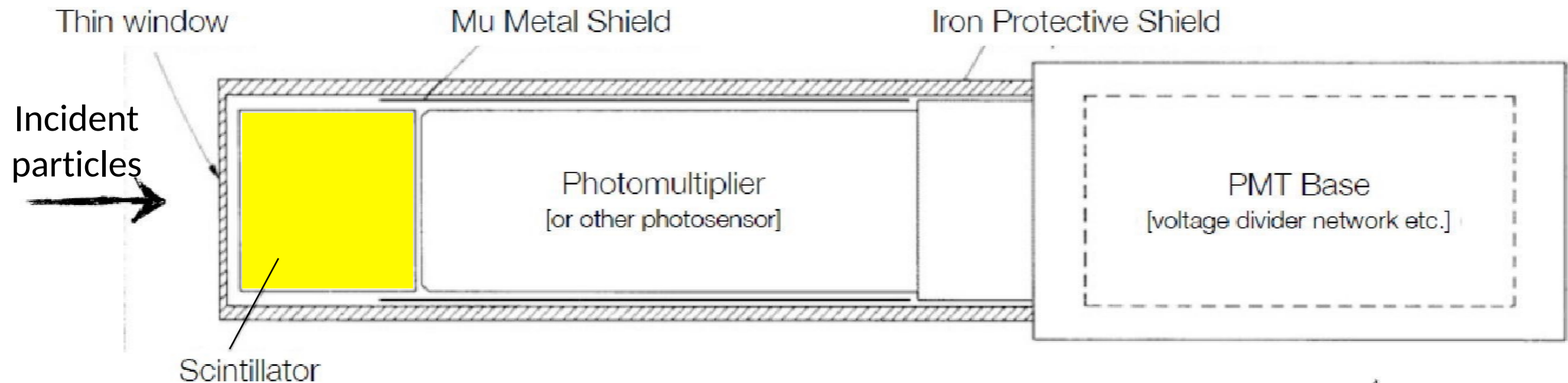
Scintillators



Minerals that emit visible light when exposed to ultraviolet light

- Basic principle:
 - A part of dE/dx is converted into visible light
 - Detected via a photosensor (eye, photomultiplier, ...)
- Properties we want in a scintillator:
 - **High efficiency** for conversion of excitation energy to prompt fluorescent radiation (vs delayed phosphorescent light)
 - **Light yield** (photons/MeV): Number of emitted photons per unit absorbed energy
 - **Linearity** between dE/dx and emitted light
 - **Transparency** to its fluorescent radiation, to allow transmission of light
 - **Frequency** of fluorescent light should be matched to photosensors
 - **Energy resolution** (in %) should be low, to measure photon energies
 - **Decay time** should be short, to avoid pile-up.
 - **Afterglow** (residual light output after the primary pulse) should be **small**
 - **Stopping power** (how much the incoming radiation is attenuated per unit thickness) should be high enough that we can capture photons with a reasonable crystal size.

Scintillators: Basic design



- **Scintillator types:**

- Inorganic crystals
- Organic scintillators
- Gases
- ... or even liquids (e.g. liquid Argon)

- **Photodetectors:**

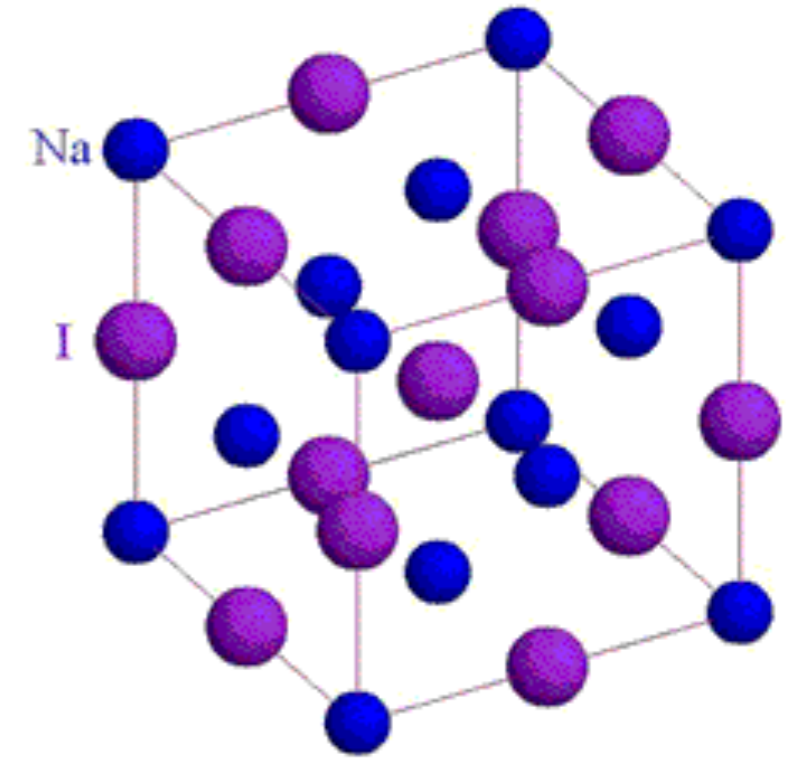
- Photomultipliers (PMTs)
- Microchannel plates (MCPs)
- Hybrid photodiodes (HPDs)
- Silicon photomultipliers

Excitation/ionization → light → photo-sensor → electric signal

Scintillators: Inorganic crystals

- Different types:

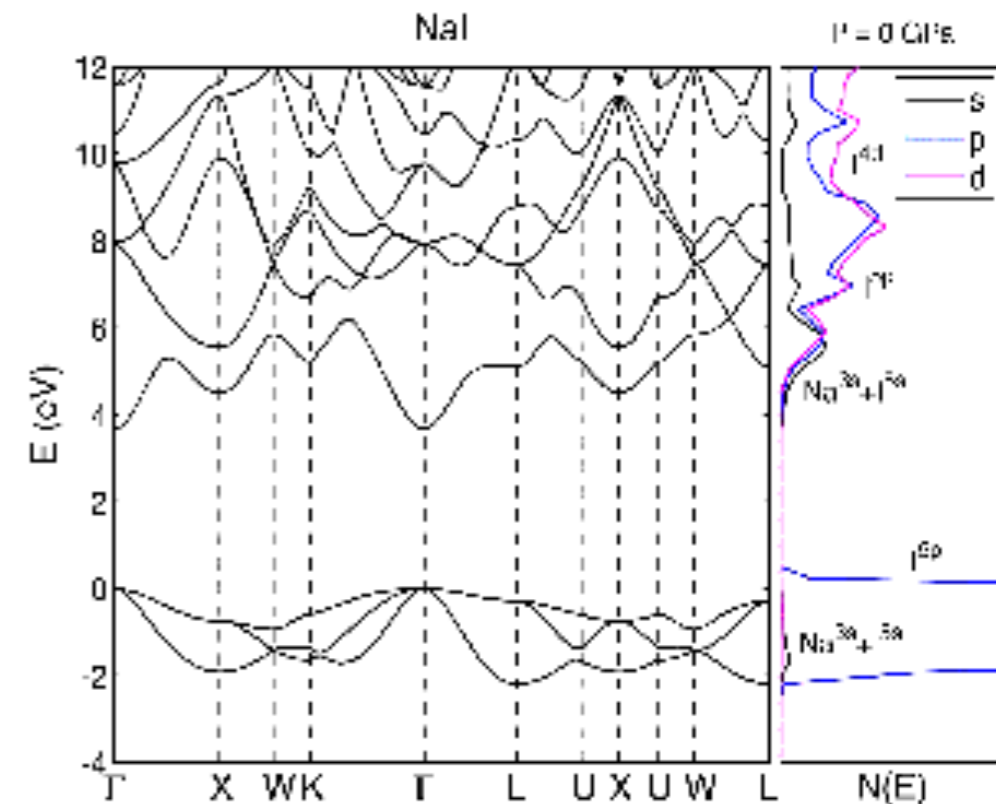
- Alkali halide: NaI(Tl), CsI(Tl), CsI(Na), LiI(Ei)
- Other slow inorganics: BGO, CdWO₄, ZnS(Ag)
- Cerium-activated fast inorganics: GSO, YAP, YAG, LSO, LuAP, LaBr₃



- First, **excitation**:

- Different bands
- Energy deposition by ionization/excitation
- Creation of electron-hole pairs
- Thermalization: all the electrons are at the bottom of the conduction band and the holes at the top of the valence band.

< 1ps



Scintillators: Inorganic crystals

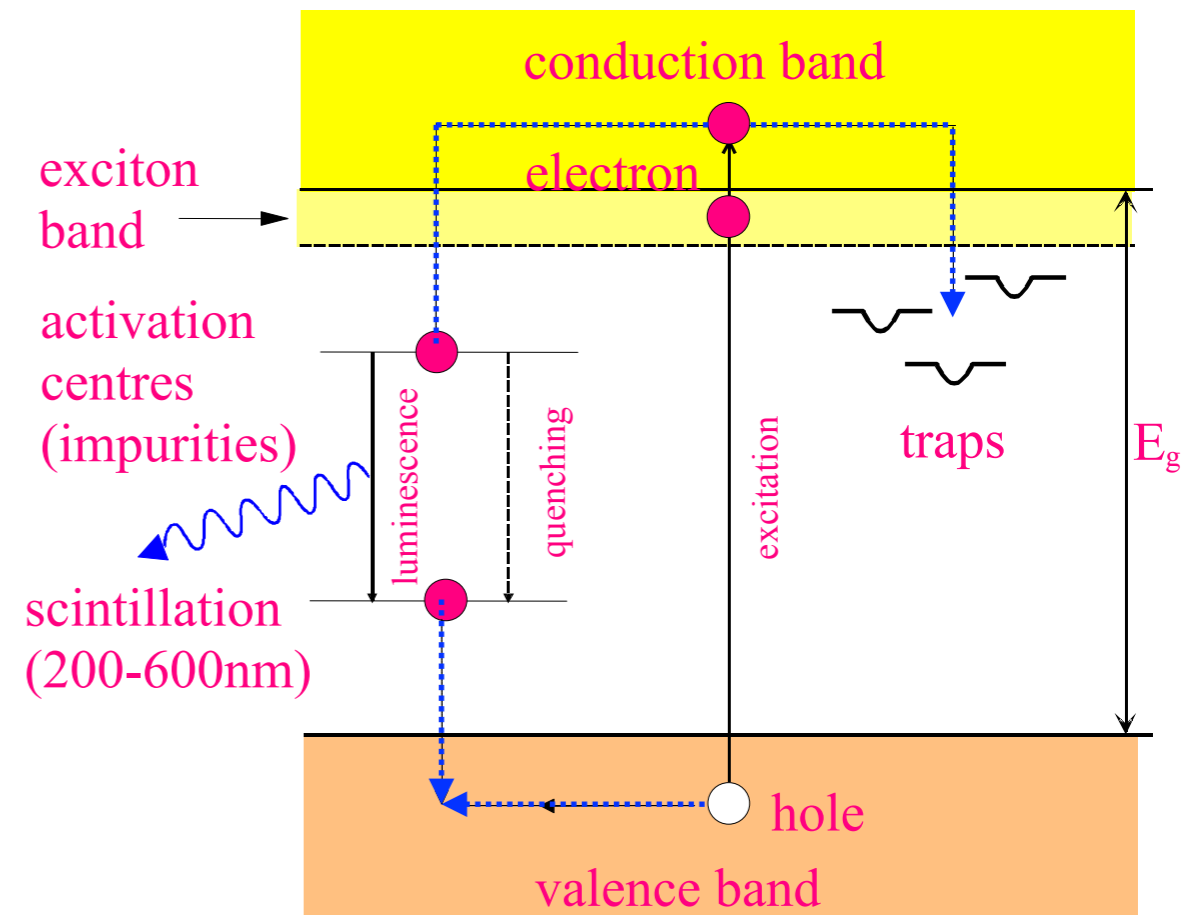
... then **de-excitation**:

- Pure crystals:
 - Emission of radiation **not efficient**, transparency problem
- Crystals with **impurities**:
 - Impurities => activation centres within the forbidden energy gap
 - After ionisation, the free electrons/holes migrate through the material and transfer their energy to the centres
 - De-excitation by photon emission (scintillation/luminescence) or by quenching (non-radiative processes in a trap)

10^{-12} to 10^{-8} s



$>10^{-10}$ s
(depending on scintillator)



CERN Academic Training Program, C. D'Ambrosio

Example: NaI(Tl)

- Doped with Tl (Thallium), typical conc. 10^{-3}
- Gap 6 eV – photon 3 eV
- 13% of deposited energy converted to light (very high light yield!)

Inorganic crystals: Properties

Parameter:	ρ	MP	X_0	R_M	dE/dx	λ_I	τ_{decay}	λ_{max}	n^*	Relative output [†]	Hygroscopic?	$d(\text{LY})/dT$ %/°C [‡]	Photons/MeV
Units:	g/cm ³	°C	cm	cm	MeV/cm	cm	ns	nm					
NaI(Tl)	3.67	651	2.59	4.8	4.8	41.4	230	410	1.85	100	yes	~0	40000
BGO	7.13	1050	1.12	2.3	9.0	21.8	300	480	2.15	9	no	-1.6	2800
BaF ₂	4.89	1280	2.06	3.4	6.6	29.9	630 ^s	300 ^s	1.50	21 ^s	no	-2 ^s	2000
							0.9 ^f	220 ^f		2.7 ^f		~0 ^f	
CsI(Tl)	4.51	621	1.85	3.5	5.6	37.0	1300	560	1.79	45	slight	0.3	
CsI(pure)	4.51	621	1.85	3.5	5.6	37.0	35 ^s	420 ^s	1.95	5.6 ^s	slight	-0.6	1100
							6 ^f	310 ^f		2.3 ^f			
PbWO ₄	8.3	1123	0.9	2.0	10.2	18	50 ^s	560 ^s	2.20	0.1 ^s	no	-1.9	200
								10 ^f		420 ^f		0.6 ^f	
LSO(Ce)	7.40	2070	1.14	2.3	9.6	21	40	420	1.82	75	no	-0.3	1400
GSO(Ce)	6.71	1950	1.37	2.4	8.9	22	600 ^s	430	1.85	3 ^s	no	-0.1	
										56 ^f		30 ^f	

* Refractive index at the wavelength of the emission maximum.

† Relative light yield measured with a bi-alkali cathode PMT.

‡ Variation of light yield with temperature evaluated at room temperature.

f = fast component, *s* = slow component

Light yield (ϵ_{sc}): fraction of energy loss going into photons

Consider a 1 MeV particle that deposits all of its energy in the scintillator:

$$\text{NaI(Tl): } \lambda_{\text{max}} = 410 \text{ nm} \Rightarrow 3 \text{ eV}$$

$$\epsilon_{\text{sc}} = \frac{(40000 \times 3 \text{ eV})}{1 \text{ MeV}} = 12\%$$

$$\text{PbWO}_4: \lambda_{\text{max}} = 560 \text{ nm} \Rightarrow 2.2 \text{ eV}$$

$$\epsilon_{\text{sc}} = \frac{(200 \times 2.2 \text{ eV})}{1 \text{ MeV}} = 0.044\%$$

$$\hbar c \approx 197.3 \text{ MeV fm}$$

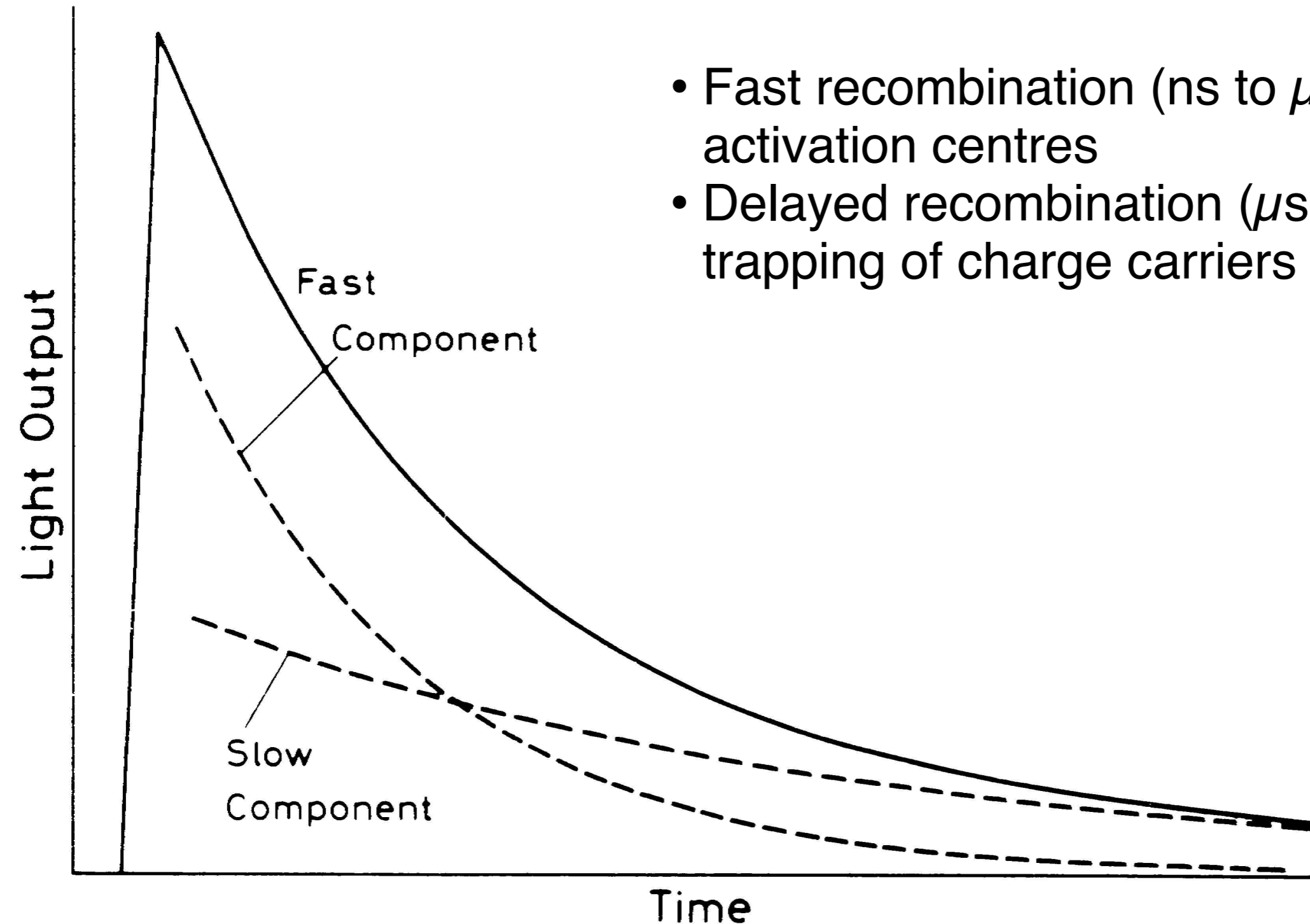
$$E_\gamma = 2\pi\hbar c/\lambda \quad 36$$

Inorganic crystals: Time response

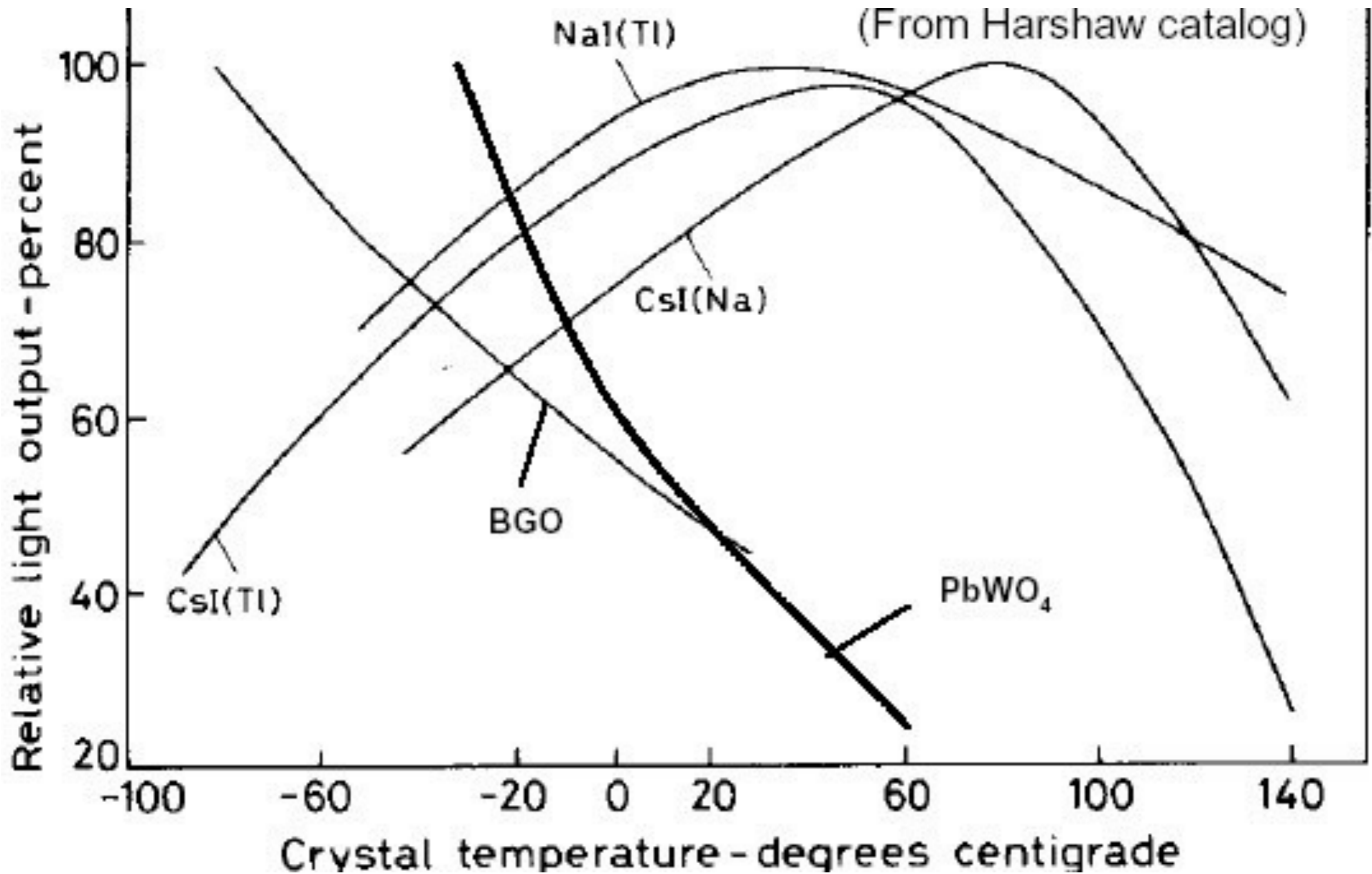
Time response usually has two components (fast & slow)

$$I(t) \propto A_f e^{-t/\tau_f} + A_s e^{-t/\tau_s}$$

- Fast recombination (ns to μ s) from activation centres
- Delayed recombination (μ s to ms) due to trapping of charge carriers at defects



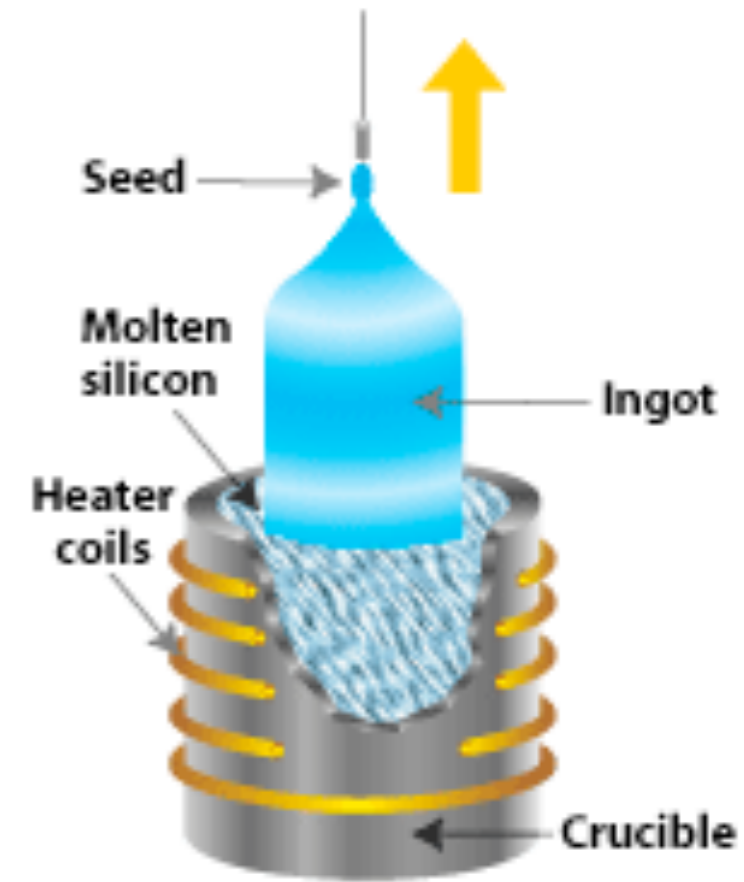
Inorganic crystals: Temperature



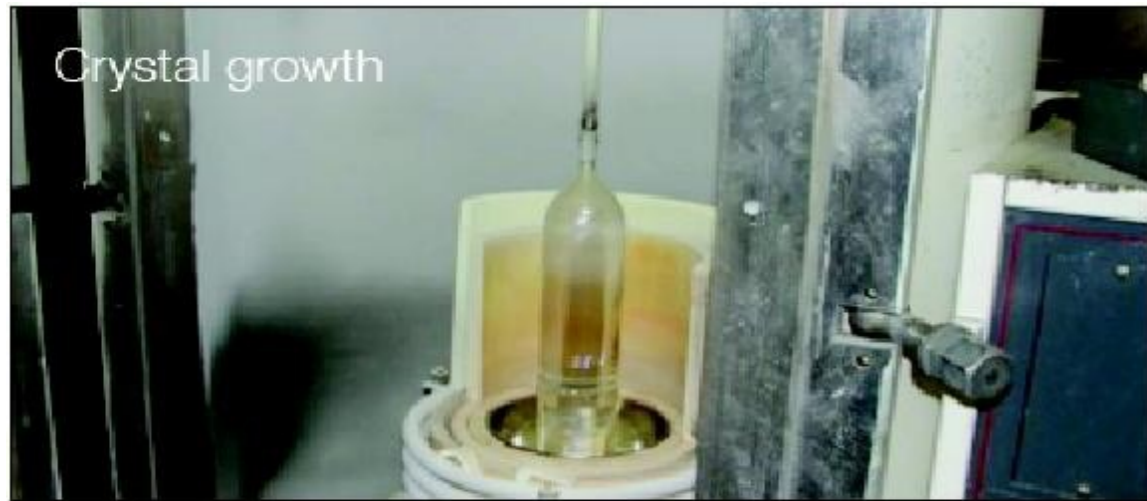
Strong temperature dependence

Inorganic crystals: CMS ECAL (example)

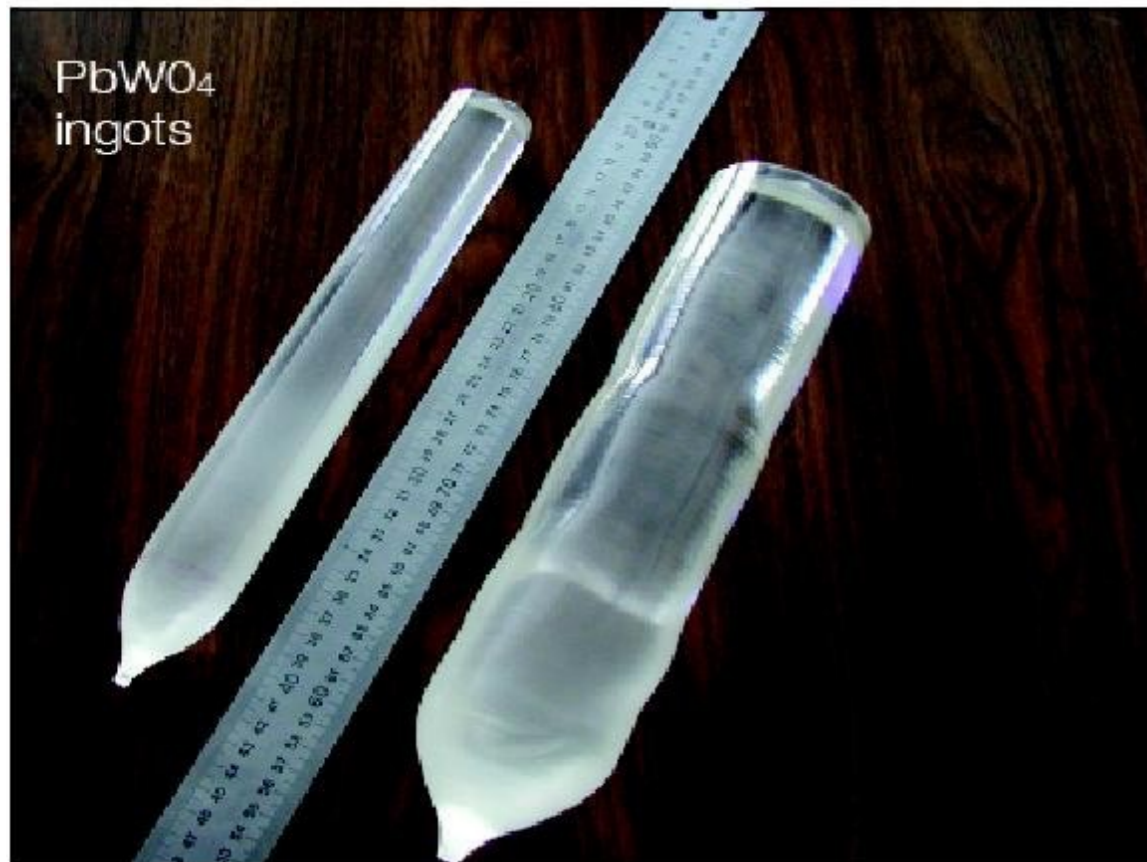
One attaches a seed crystal to the bottom of a vertical arm such that the seed is barely in contact with the material at the surface of the melt. The arm is raised slowly, and a crystal grows underneath at the interface between the crystal and the melt.



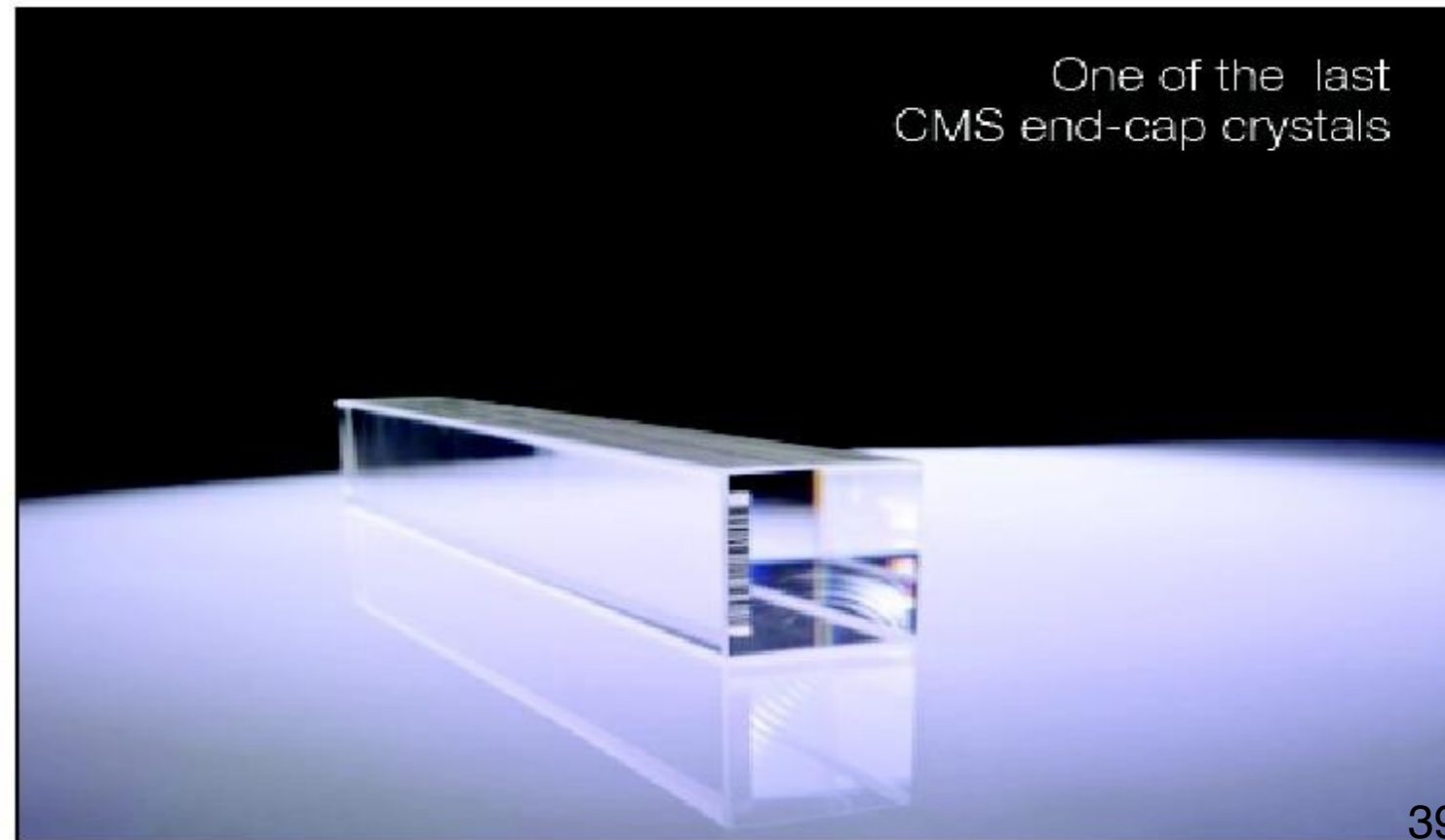
Crystal growth



PbWO₄ ingots



One of the last CMS end-cap crystals



Scintillators: Liquid noble gases

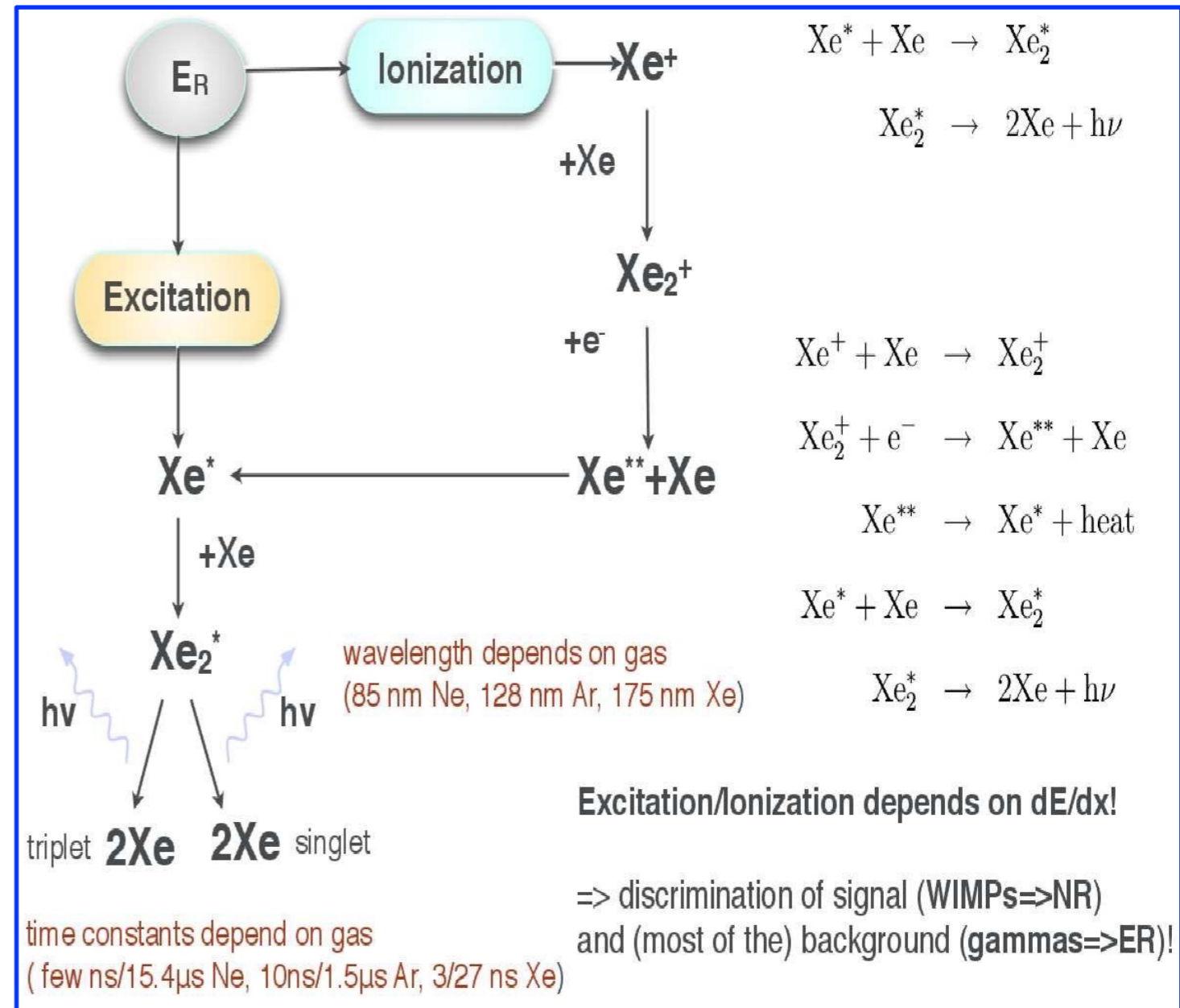
- Several different types:

- Helium (He)
- Liquid Neon (LNe)
- Liquid Argon (LAr)
- Liquid Xenon (LXe)
- ...

- Main features:

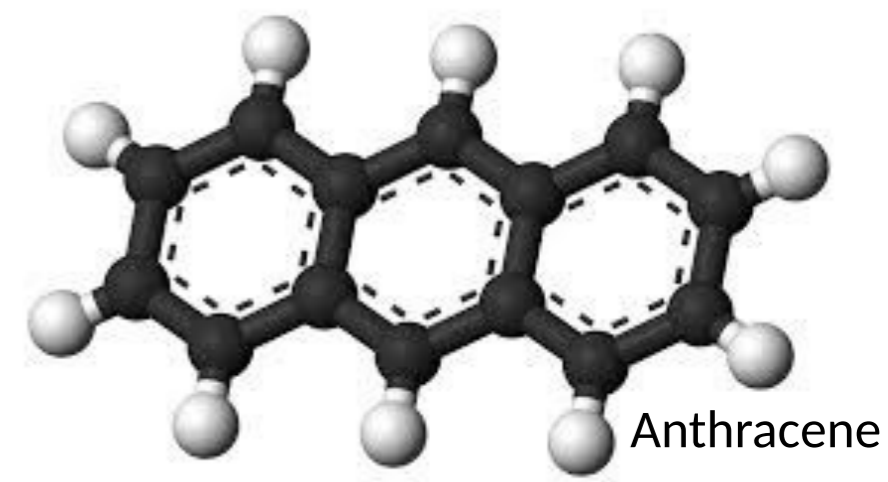
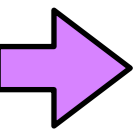
- High scintillation light yields
- Transparent to the scintillation light
- Large detector masses are feasible
- Can be made very pure (important for eliminating radiological background in low-signal searches)

Example: use of LXe in WIMP searches



NR = Nuclear recoil ; ER = Electronic recoil

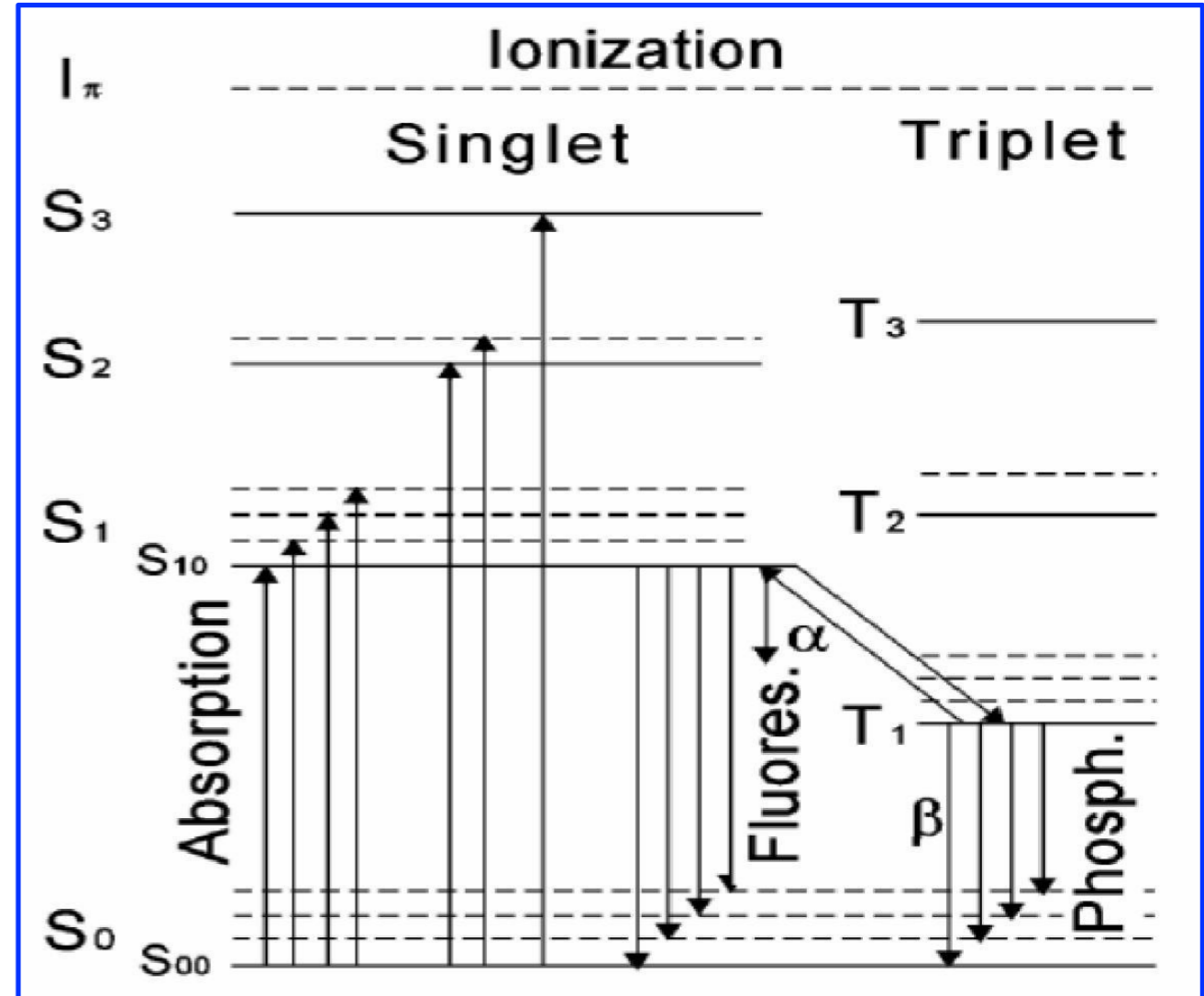
Scintillators: Organic



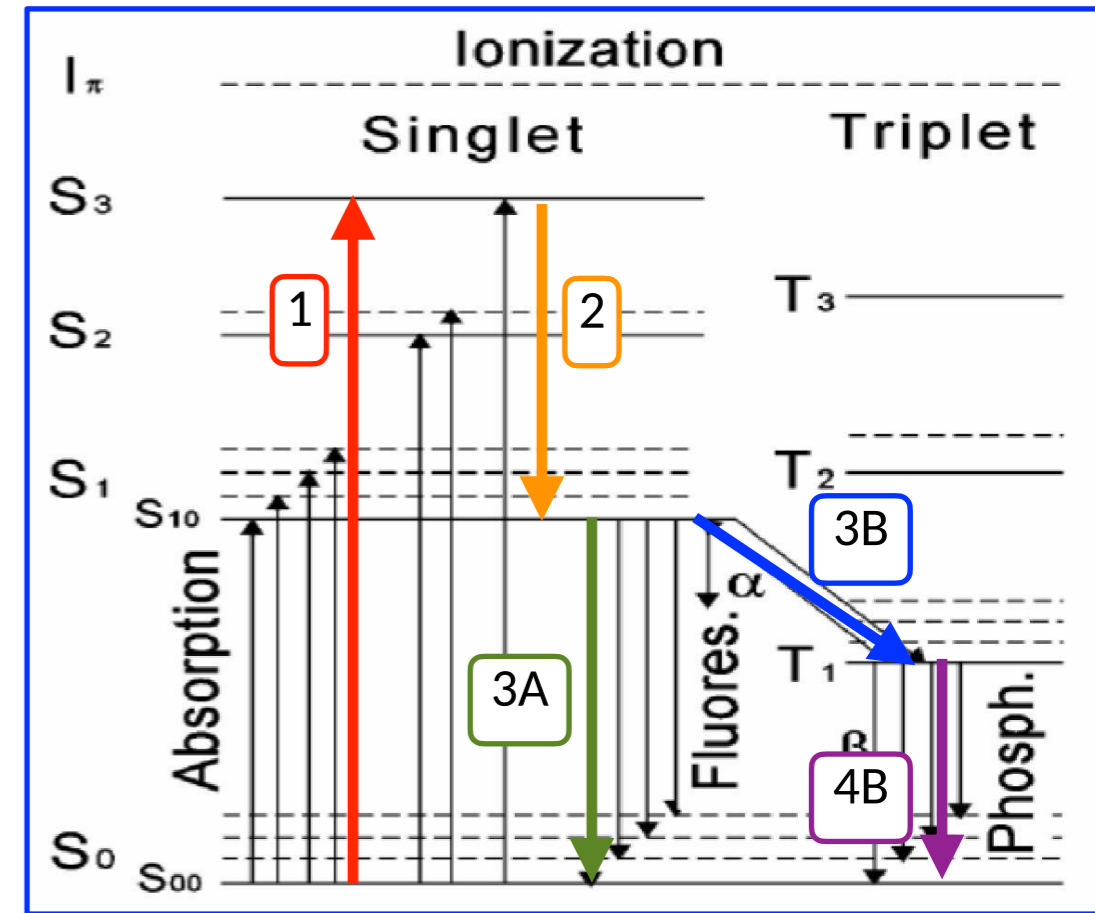
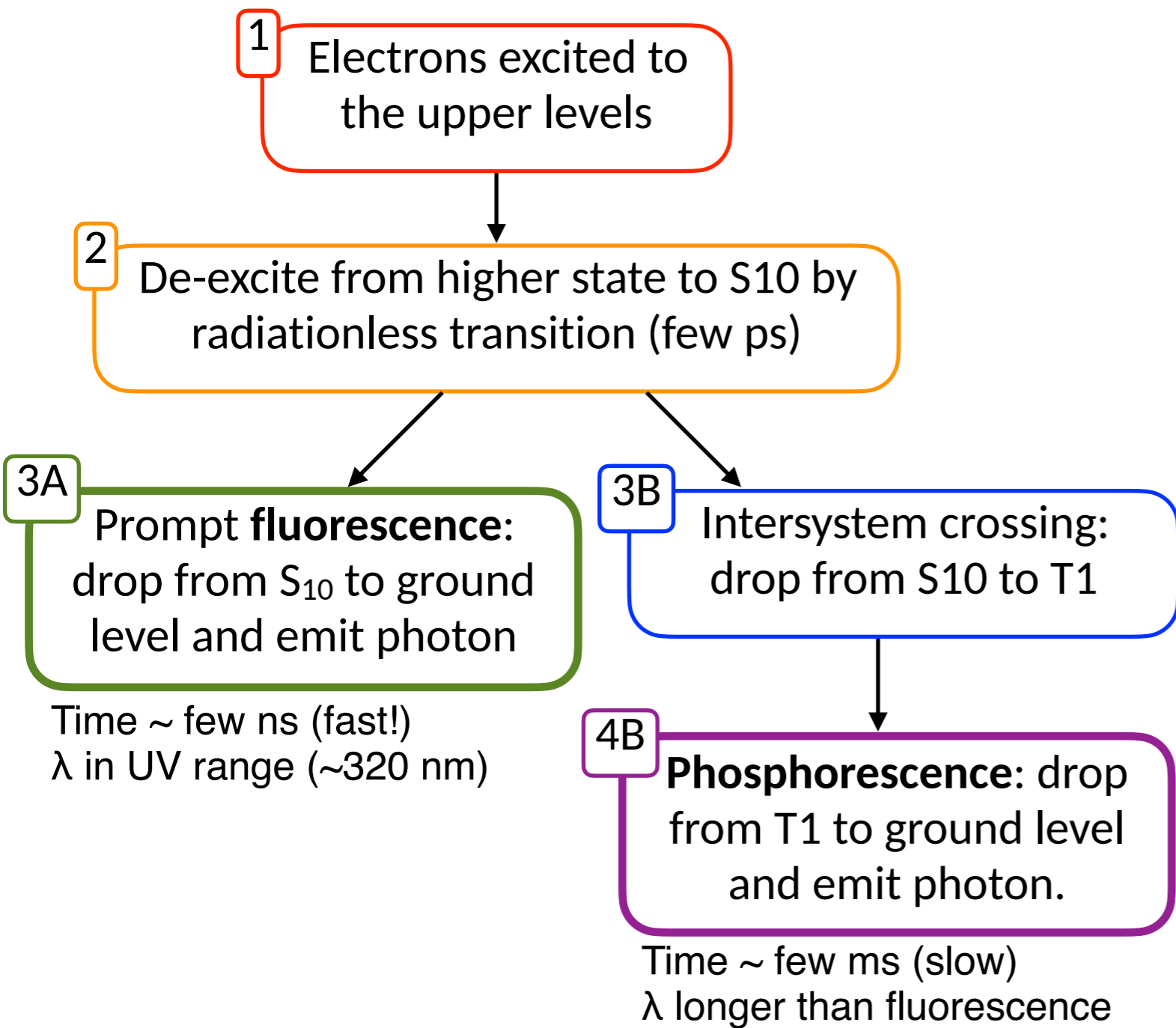
- Based on aromatic hydrocarbon compounds:
 - Organic -- composed of { C, H, O, N }
 - Examples: Anthracene [C₁₄H₁₀], Stilbene [C₁₄H₁₂], ...
 - Scintillation occurs at the level of a single molecule, so ~ independent of the physical state

Mechanism

- Delocalized electrons in π -orbitals
- Light emitted in transitions between energy levels
- S = singlet states (spin 0)
- T = triplet states (spin 1)
- Fine structure: each S/T state split into additional levels



Scintillators: Organic

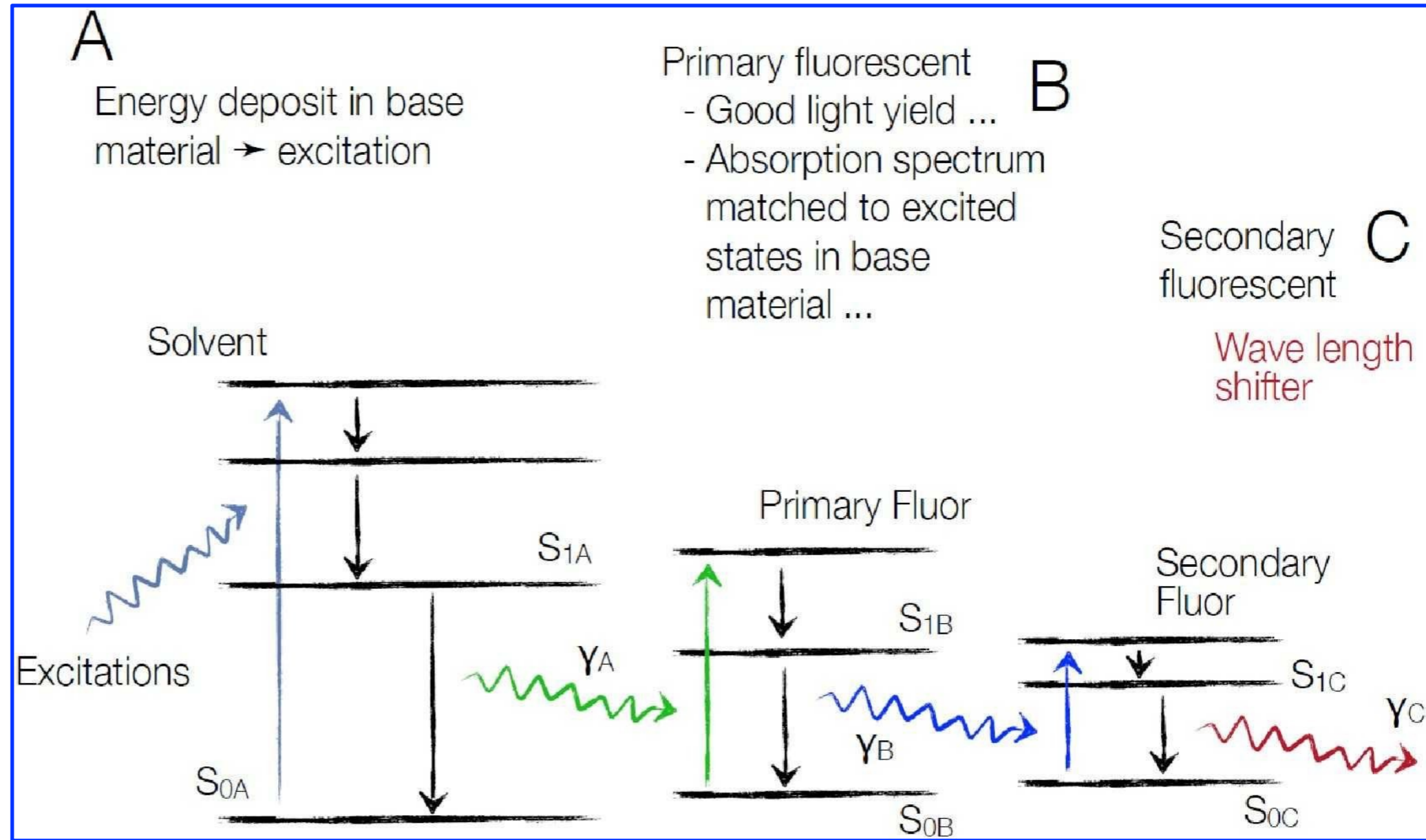


- Problems:
- Light in UV range (harder to detect) => Use wavelength shifters (WLS)
 - Prompt fluorescence yield low

Possible to distinguish fluorescence and phosphorescence based on timing, wavelength.

Scintillators: Organic: Wavelength shifting

From HansChristian SchultzCoulon
KirchhoffInstitut für Physik

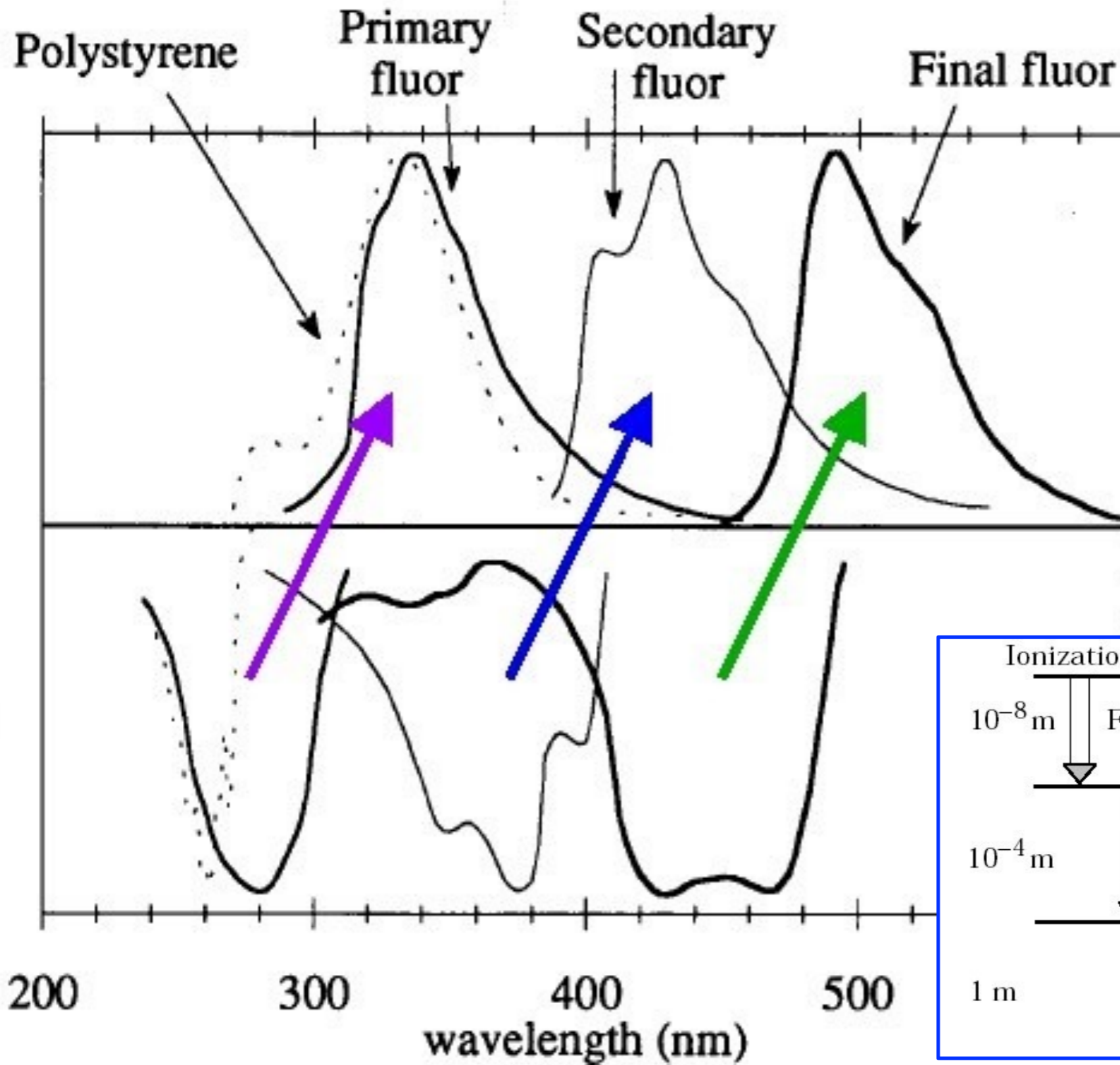


In practice:

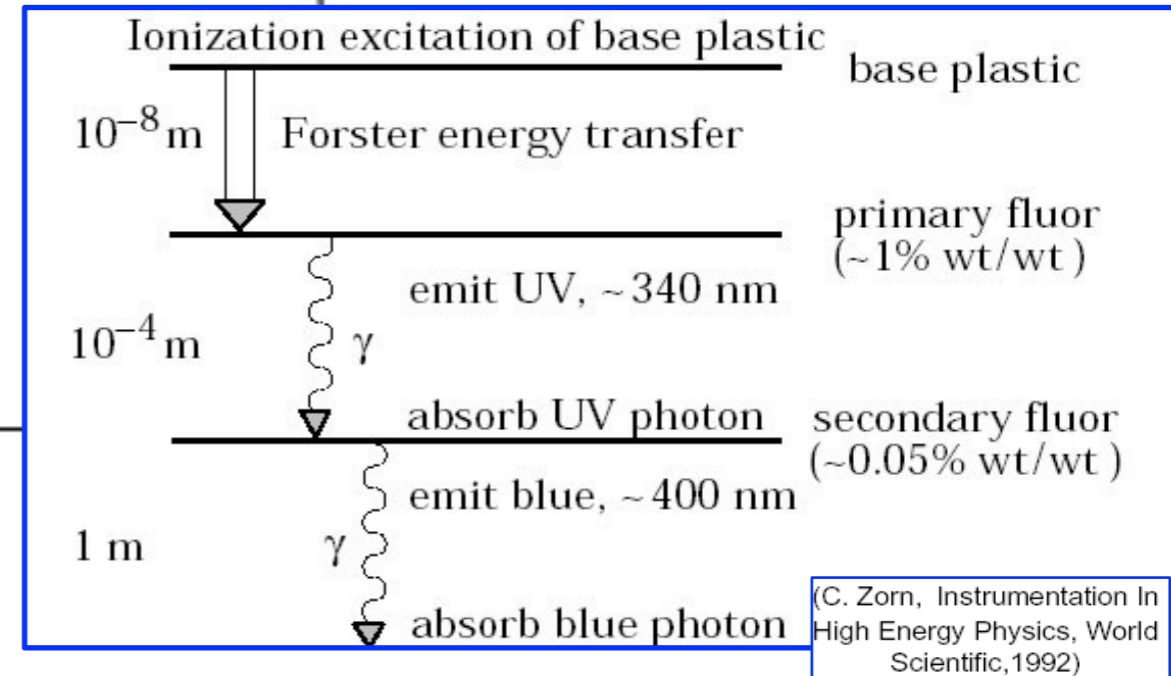
- Solution of organic scintillators [**dissolved in plastic or liquid**]
- + large concentration of primary 'fluor' (perhaps 1-3%)
- + smaller concentration of secondary 'fluor'
- + maybe more

⇒ step down the photon energy until λ is well matched to photodetector.
[Also helps avoid re-absorption of light]

Scintillators: Organic: Wavelength shifting



Adapts light to spectral range of photosensor



(C. Zorn, Instrumentation In High Energy Physics, World Scientific, 1992)

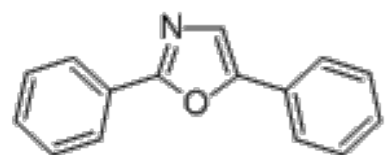
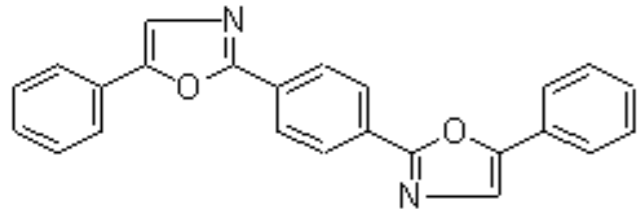
Emission spectrum for step (n) and absorption spectrum for step (n+1) must overlap.

Scintillators: Organic: Composition

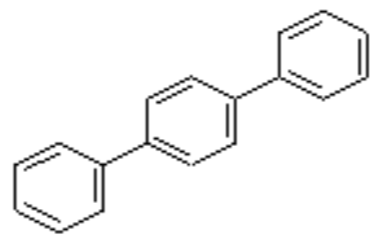
Some widely used solvents and solutes:

State	Solvent	Primary fluor	Secondary fluor
Liquid	Benzene	p-terphenyl	POPOP
	Toluene	DPO	BBO
	Xylene	PBD	BPO
Plastic	Polyvinylbenzene	p-terphenyl	POPOP
	Polyvinyltoluene	DPO	TBP
	Polystyrene	PBD	BBO or DPS

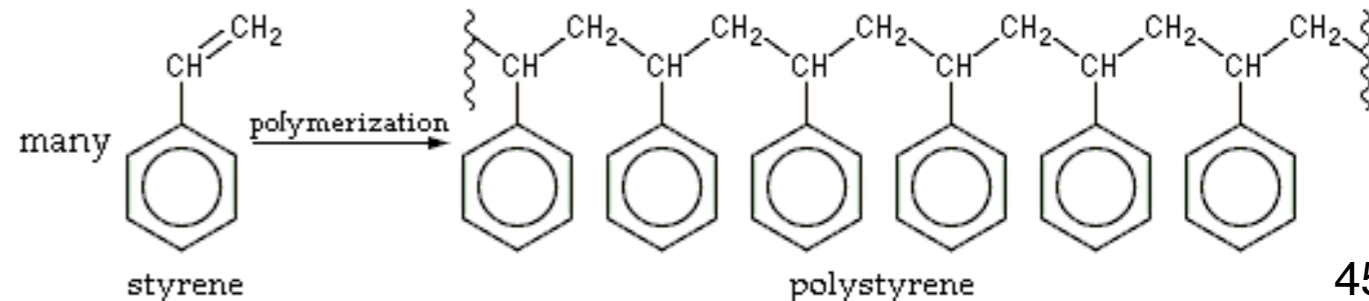
POPOP



DPO

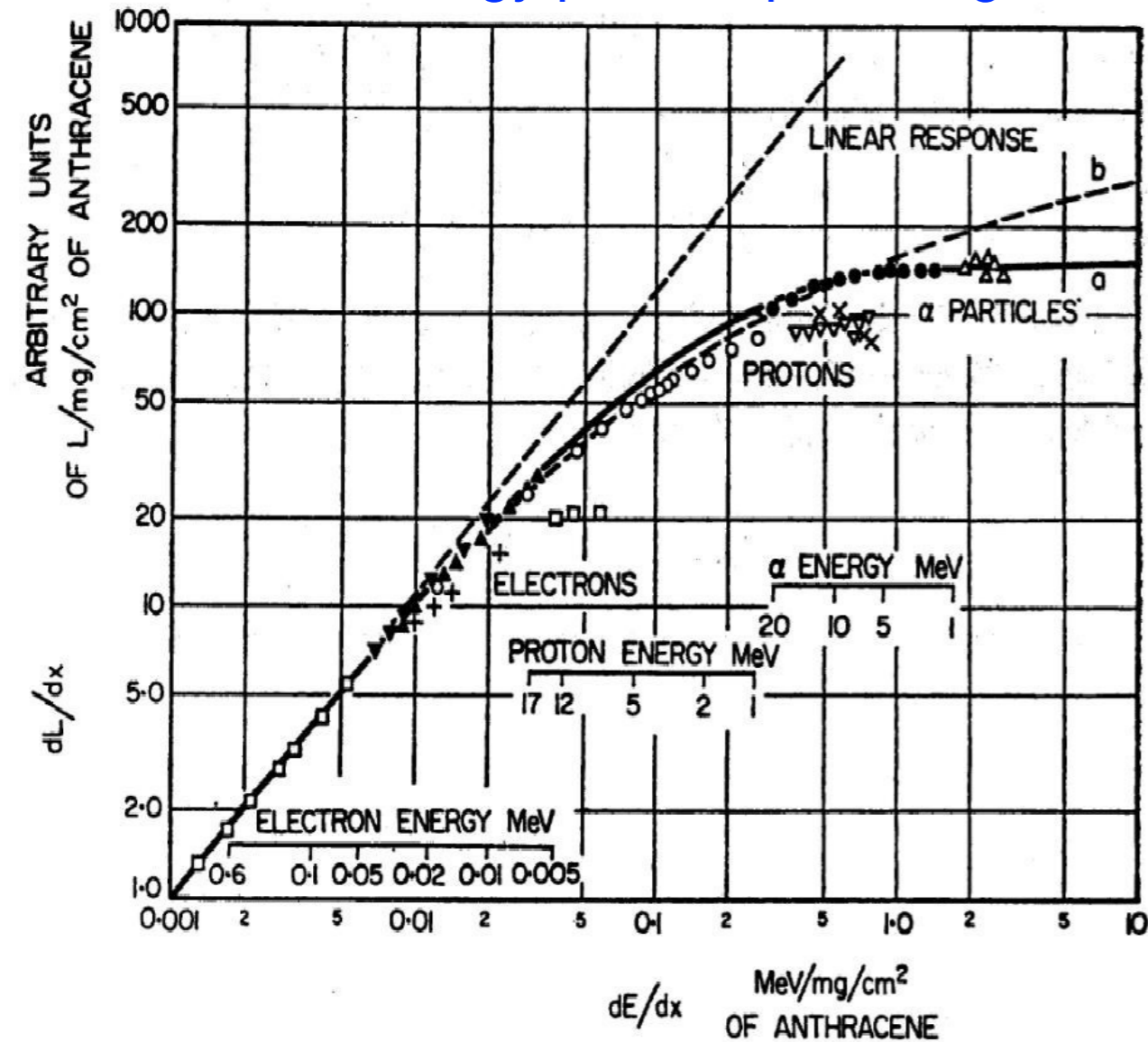


p-terphenyl



Scintillators: Organic: Light output

Fluorescent energy per unit path length dL/dx

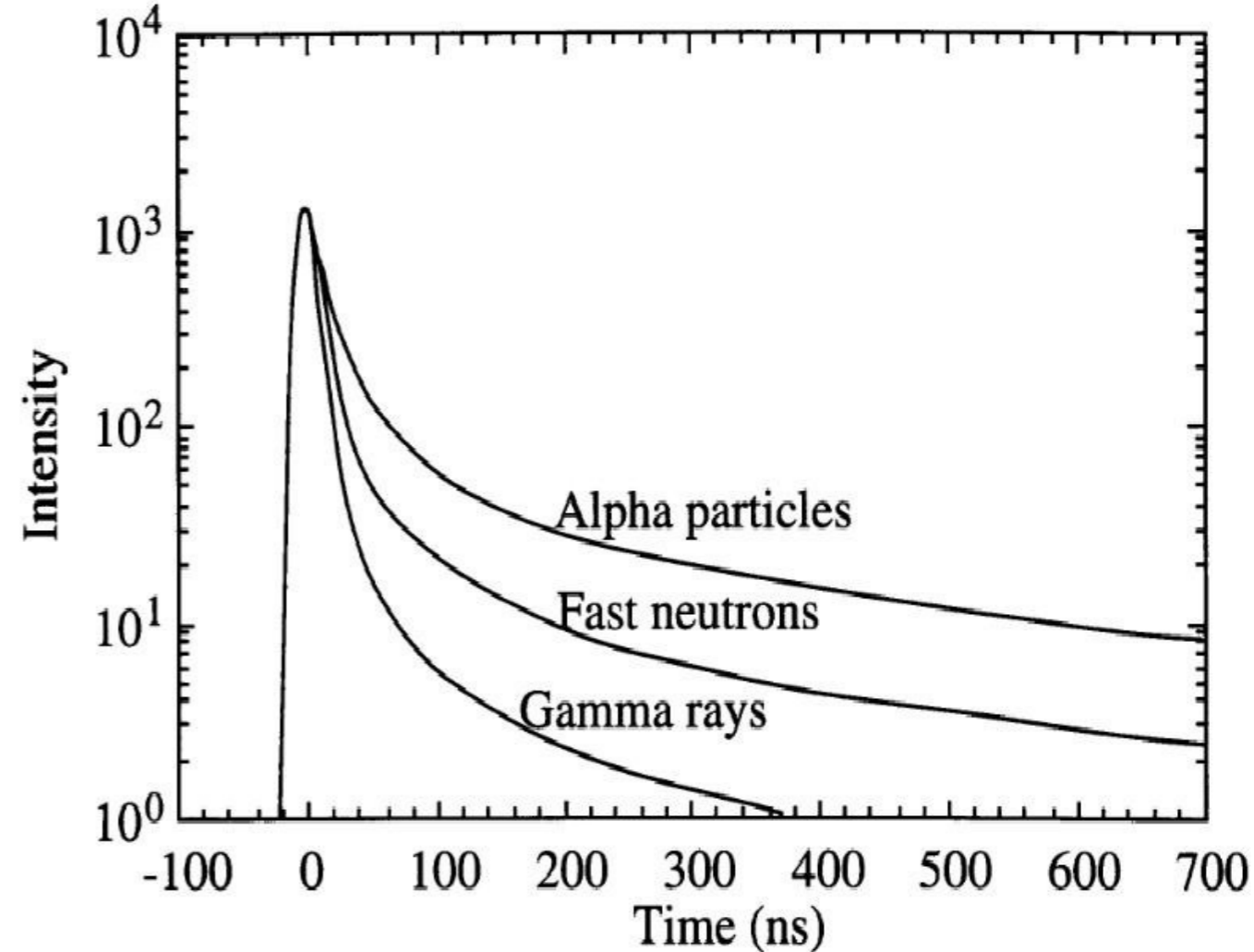


Variation of specific fluorescence dL/dx in anthracene with specific energy loss dE/dx (Brooks, from Birks)

Quenching: nonlinear response due to saturation of available states.

$$\frac{dL}{dx} = L_0 \frac{\frac{dE}{dx}}{1 + k_B \frac{dE}{dx}}$$

Light output vs time for different particles



Discrimination between particles: time response varies according to incoming particle type.

(Caveat: may be affected by ageing, magnetic field effect, damage radiation, ...)

Scintillators: Organic: Properties

Scintillator material	Density [g/cm ³]	Refractive Index	Wavelength [nm] for max. emission	Decay time constant [ns]	Photons/MeV
Naphtalene	1.15	1.58	348	11	$4 \cdot 10^3$
Antracene	1.25	1.59	448	30	$4 \cdot 10^4$
p-Terphenyl	1.23	1.65	391	6-12	$1.2 \cdot 10^4$
NE102*	1.03	1.58	425	2.5	$2.5 \cdot 10^4$
NE104*	1.03	1.58	405	1.8	$2.4 \cdot 10^4$
NE110*	1.03	1.58	437	3.3	$2.4 \cdot 10^4$
NE111*	1.03	1.58	370	1.7	$2.3 \cdot 10^4$
BC400**	1.03	1.58	423	2.4	$2.5 \cdot 10^2$
BC428**	1.03	1.58	480	12.5	$2.2 \cdot 10^4$
BC443**	1.05	1.58	425	2.2	$2.4 \cdot 10^4$

* Nuclear Enterprises, U.K.
 ** Bicron Corporation, USA

c.f. NaI(Tl): $\sim 40k$ photons / MeV

Scintillators: Pulse shape discrimination

- For multiple types of scintillator, we saw that time response can include fast and slow components
 - e.g. inorganic crystals like CsI(Tl)
 - e.g. organic scintillators like Stilbene
- We also saw that the mechanisms available can be influenced by nonlinear/saturation/quenching effects.
 - i.e. a large local energy deposit can exhaust all of the available states for a certain transition pathway.
 - Details are different for organic vs inorganic
- Some particles deposit energy faster
 - e.g. electrons vs muons vs alpha ($z=2$)
 - e.g. neutrons vs photons
- Consequence: time response can depend on particle type; can infer PID information from pulse shape.
- Not much used in high-rate, high-energy detectors today, but can be useful in low-rate or nuclear experiments.

Scintillators: Comparison

	Advantages	Disadvantages	Used for
Inorganic	<p>High light yield (~ 0.13)</p> <p>High density</p> <p>Good energy resolution</p>	<p>Complicated crystal growth</p>	<p>Calorimetry</p> <p>Gamma spectroscopy</p> <p>Charged particle detection</p> <p>...</p>
Organic	<p>Very fast</p> <p>Easily shaped</p>	<p>Lower light yield (~ 0.03)</p> <p>Radiation damage (esp. for plastics; less for liquids)</p> <p>Aging</p>	<p>Time measurement</p> <p>Particle discrimination</p> <p>Charged particle detection</p> <p>...</p>

"Typical" advantages & disadvantages; properties of individual materials vary a lot.

Getting the light to a photodetector

Light must be guided to photosensor with minimal losses
=> use Total Internal Reflection, requires careful optimization of geometry.



Saint-Gobain:

Typical light pipe geometries include:

- **Right Cylinders:** used when the light pipe diameter is the same as the scintillator diameter.
- **Tapered Cones:** transition pieces between square-to-round or round-to-round cross-sections
- **Fish Tail:** Transition pieces from thin, rectangular cross-sections to round cross-sections
- **Adiabatic:** provide the most uniform light transmission from the scintillator exit end to the PMT; the cross-sectional areas of the input and PMT faces are equal

Next time: how to detect and measure the light once it reaches the photodetector

... and we'll come back to interaction of other neutral particles with matter.

Some useful further resources for scintillator info: Leo; PDG review "[Particle detectors at accelerators](#)" sections 35.3 + 35.4 + maybe 35.7.

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Particules
Astroparticules
Cosmologie*

Master 2 Recherche

Bruno Mazoyer - LAT Orsay

Example exam questions from previous years

Note: The exam format has changed over time, and the syllabus has also evolved somewhat. The point is not to give you the exact style or content of this year's questions, but to help you prepare.

Exam questions

From 2016:

We consider photons in the energy range $10 \text{ eV} < E_\gamma < 100 \text{ GeV}$. For interactions with matter, the dominant processes in this region are the Compton effect, pair production, and the photoelectric effect.

- (a) For each of these three processes, draw a Feynman diagram.
- (b) Sketch a graph showing how the interaction cross-section of each of these three processes varies with E_γ on a log-log scale for a material of intermediate Z , labelling any important features.

Exam questions

From 2017:

A Higgs boson is produced inside a detector and decays to a pair of photons. Each of the photons enters an electromagnetic calorimeter, producing a shower.

- (a) Explain briefly how the shower develops, in terms of the key physical processes involved.
- (b) How do the dimensions of the shower relate to the radiation length X_0 and the critical energy E_c of the material?

Exam questions

From 2021:

Q3 (approx. 10–15 min)

We wish to study photons of energy E_γ between 10 eV and 10 keV. We use a thin plate of material.

- (a) What is the most probable way for the photons to interact with the material? Identify the process and draw a Feynman diagram (including all participating particles).
- (b) We have plates of various materials available (e.g. Si, Cu, Fe, Pb, ...), and plan to study how the interaction probability varies between them. What property of the *material* has the most influence on the interaction probability? Roughly how does the cross-section scale with this property? (The question refers to the physical properties of the material, not the dimensions of the plate.)
- (c) We observe an event in which an incoming 18.0 keV photon produces a single electron of energy 9.0 keV and no other outgoing particles. What can we deduce from this? (The value of 9.0 keV represents the initial energy of the electron, ignoring any subsequent energy loss in the material.)
- (d) For a particular material, we measure how the cross-section varies as a function of photon energy. We notice that the variation is not smooth but has a number of ridges/peaks in the plot of cross-section vs energy. Suggest a physical origin for these ridges.